

# Student Exploration: Limiting Reactants (ANSWER KEY)

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### Student Exploration: Limiting Reactants

**Vocabulary:** chemical equation, chemical formula, chemical reaction, coefficient, limiting reactant, molecule, product, reactant, subscript

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

How many total hot dogs can you make? \_\_\_\_\_

Which ingredient limited the number of hot dogs you could make? \_\_ \_\_\_\_\_

Which ingredient will you have leftovers of? \_\_\_\_\_

### Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants Gizmo™*, you can determine which reactant is limiting in various scenarios.

To begin, make sure  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$  is selected. The small "2" in  $\text{H}_2$ ,  $\text{O}_2$ , and  $\text{H}_2\text{O}$  is a **subscript**. Subscripts represent the number of atoms in a **molecule**.

Use the sliders to set the number of  $\text{O}_2$  molecules and  $\text{H}_2$  molecules to two.

How many hydrogen molecules ( $\text{H}_2$ ) are there? \_\_\_\_\_ Hydrogen atoms? \_\_\_\_\_

How many oxygen molecules ( $\text{O}_2$ ) are there? \_\_\_\_\_ Oxygen atoms? \_\_\_\_\_

How many  $\text{H}_2\text{O}$  molecules do you think will form when these four molecules react? \_\_\_\_\_

Click **Play** (). How many  $\text{H}_2\text{O}$  molecules actually formed? \_\_\_\_\_

<b>Activity A:</b>  <b>Water reaction</b>	<b>Get the Gizmo ready:</b> <ul style="list-style-type: none"><li>• Make sure <math>\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}</math> is selected.</li><li>• Click <b>Reset</b> ().</li></ul>	
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# Limiting Reactant Student Exploration Sheet Answer

**Léo Pomar**



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