Limiting Reactants and Percent Yield Worksheet

Name_____ Date_____ Period_____

CaCO₂ → CaO + CO₂

Calculate the theoretical yield of Calcium Oxide if 24.8 grams of Calcium Carbonate decomposes. Calculate the percent yield if you got 13.1 grams in the laboratory.

2) Al + CuSO₄ → Cu + Al₂(SO₄)₃

What is the percent yield if you experimentally produce 3.65 grams of copper when 1.87 grams of Aluminum reacts with 9.65 grams of Copper (II) Sulfate?

3) SO₂ + 2NaOH → Na₂SO₃ + H₂O

There are 100.0 grams of each reactant available. Determine the limiting reactant in this equation.

Limiting Reactant And Percent Yield Answer Key

Ralph Thomas Myers

Limiting Reactant And Percent Yield Answer Key:

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