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 $\text{aq C}_3\text{H}_7\text{NO}_2 + \text{aq H}_3\text{O}^+ \rightleftharpoons \text{C}_3\text{H}_8\text{NO}_2^+ + \text{aq H}_2\text{O}$ $\text{l CH}_3\text{O}_2\text{H} + \text{aq NH}_3 \rightleftharpoons \text{aq CH}_3\text{CO}_2^- + \text{aq NH}_4^+$
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web the ionization of weak acids and weak bases many acids and bases are weak that is they do not ionize fully in aqueous solution a solution of a weak acid in water is a mixture of the nonionized acid hydronium ion and the conjugate base of the acid with the nonionized acid present in the greatest concentration

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web oct 10 2022 correct answer d salt water explanation neutralization is a chemical reaction between an acid and a base resulting in the formation of a salt and water the product of neutralization is therefore salt water as the acid and base combine to form a salt compound and water molecules rate this question 4