The kinetic theory of matter

Arrangement of Matter	State of Matter	Arrangement and Movement of Particles
888	SOLID Has a fixed volume and shape Cannot be compressed	 The particles are packed closely together in an orderly manner. There are strong forces between the particles. The particles can only vibrate and rotate about their fixed positions.
£	LIQUID Has fixed volume but does not have fixed shape (takes the shape of container) Cannot be compressed easily	 The particles are packed closely together but not in orderly arrangement. Held together by strong forces but weaker than the forces in a solid. The particles can vibrate, rotate and move throughout the liquid. They collide against each other.
000	GAS Does not have a fixed shape or volume Can be compressed easily	 The particles are very far apart from each other and in a random motion. Weak forces between the particles. The particles can vibrate, rotate and move freely. The rate of collision is greater than the rate of collision in a liquid.

<u>Kinetic Molecular Theory Reinforcement Worksheet</u> <u>Answers</u>

Charles Proteus Steinmetz, American Electrochemical Society. Meeting

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Kinetic Molecular Theory Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908*

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