



Lancer Gts 2009 User Guide

F Rizvi



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arrhenius model of bases a substance that contains a hydroxide group and dissociates to produce a hydroxide

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the OH^- basic solution one in which the H^+ is less than the OH^- alkaline solutions the H^+ of a basic solution is less than $1 \times 10^{-7} \text{ M}$

they are the same as basic solutions

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thus ruling out that S^{2-} is the weakest base

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compounds have characteristics of both acids and bases in the same molecule we will explore the relationship between

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molecule that gains a hydrogen ion becomes this which is positively charged what did lewis propose an acid accepts a pair of

electrons during a reaction while a base donates a pair of electrons

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