



Equations

Name _____

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Write a balanced chemical equation to represent each of the following chemical reactions.

1. iron + sulfur \rightarrow iron(II) sulfide
2. zinc + copper(II) sulfate \rightarrow zinc sulfate + copper
3. silver nitrate + sodium bromide \rightarrow sodium nitrate + silver bromide
4. potassium chlorate (heated) \rightarrow potassium chloride + oxygen
5. water (electricity) \rightarrow hydrogen + oxygen
6. mercury(II) oxide (heated) \rightarrow mercury + oxygen
7. potassium iodide + lead(II) nitrate \rightarrow lead(II) iodide + potassium nitrate
8. aluminum + oxygen \rightarrow aluminum oxide
9. magnesium chloride + ammonium nitrate \rightarrow magnesium nitrate + ammonium chloride
10. iron(III) chloride + ammonium hydroxide \rightarrow iron(III) hydroxide + ammonium chloride
11. sodium peroxide + water \rightarrow sodium hydroxide + oxygen
12. iron(III) oxide + carbon \rightarrow iron + carbon monoxide
13. iron + water \rightarrow hydrogen + iron(III) oxide
14. iron(III) chloride + potassium hydroxide \rightarrow potassium chloride + iron(III) hydroxide
15. aluminum + sulfuric acid \rightarrow aluminum sulfate + hydrogen
16. sodium carbonate + calcium hydroxide \rightarrow sodium hydroxide + calcium carbonate



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Table of Contents J Weston Walch Publisher Answer Key

1. Understanding the eBook J Weston Walch Publisher Answer Key
 - The Rise of Digital Reading J Weston Walch Publisher Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying J Weston Walch Publisher Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an J Weston Walch Publisher Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from J Weston Walch Publisher Answer Key
 - Personalized Recommendations
 - J Weston Walch Publisher Answer Key User Reviews and Ratings
 - J Weston Walch Publisher Answer Key and Bestseller Lists
5. Accessing J Weston Walch Publisher Answer Key Free and Paid eBooks

- J Weston Walch Publisher Answer Key Public Domain eBooks
- J Weston Walch Publisher Answer Key eBook Subscription Services
- J Weston Walch Publisher Answer Key Budget-Friendly Options
- 6. Navigating J Weston Walch Publisher Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - J Weston Walch Publisher Answer Key Compatibility with Devices
 - J Weston Walch Publisher Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of J Weston Walch Publisher Answer Key
 - Highlighting and Note-Taking J Weston Walch Publisher Answer Key
 - Interactive Elements J Weston Walch Publisher Answer Key
- 8. Staying Engaged with J Weston Walch Publisher Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers J Weston Walch Publisher Answer Key
- 9. Balancing eBooks and Physical Books J Weston Walch Publisher Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection J Weston Walch Publisher Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine J Weston Walch Publisher Answer Key
 - Setting Reading Goals J Weston Walch Publisher Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of J Weston Walch Publisher Answer Key
 - Fact-Checking eBook Content of J Weston Walch Publisher Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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