(ii)	What would you observe during reaction 1?
	[1]
(III)	In reaction 1, compound E was heated under reflux with excess K2Cr2O/H2SO4.
	Suggest why these conditions were used, rather than the reaction mixture being distilled during the process.
	[1]

[1]

- (iv) Name the type of reaction taking place in reaction 2.
- (c) Compound E is a single stereoisomer.
 - (i) Draw the skeletal formula of one other stereoisomer of compound E and state the type of stereoisomerism.

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