

CHEMISTRY 2023

Max. Marks: 60

Time: 3 Hour

SECTION "A" MULTIPLE CHOICE QUESTIONS (MCQs)

Marks: 12

NOTE: (i) Attempt all the Questions of this Section.

(ii) Do not copy down the part questions. Write only the answer against the proper number of the Question and its part according to the question Paper.

(iii) Each question carries 1 Marks

1. Choose the correct Answer for each from the given question:

1. The system will be stable in equilibrium when:

* $K_c = Q_c$ ✓ * $K < Q_c$ * $K > Q_c$ * none of these

2. Corrosive effects on skin is caused by:

* acid ✓ * base * salt * all of these

3. If pH value is greater than 7, then solution will be:

* Acidic * Basic ✓ * Natural * Amphoteric

4. The functional group $-COOH$ is used for:

* Alkyne * Alcohol * Phenol * Carboxylic acid ✓

5. The branch of Chemistry which deals with the hydrocarbon and their derivatives, is known as:

* Organic Chemistry ✓ * Inorganic Chemistry

* Bio Chemistry * Physical Chemistry

6. Glucose is:

* vitamin * protein * lipid * carbohydrate ✓

7. Amino acids are building block of:

* nucleic acid * protein ✓ * vitamin * lipids

8. The centrifuge machine is used for the separation of:

* juice * pH * mud * crystal ✓

9. The altitude of stratosphere is:

* 40 to 45 km * 50 to 55 km ✓ * 60 to 65 km * 70 to 75 km

10. Ozone is a gas that is found in the layer of:

* Troposphere * Mesosphere

* Stratosphere ✓ * Exosphere

11. Fresh water is present on the surface of earth:

* 0.3% * 3% ✓ * 0.2% * 2%

12. The citric acid is used in the preparation of cold drinks for

* sweet taste * bitter taste

* sour taste ✓ * salty taste

June 2013 Ocr Chemistry Past Paper F332

Dave Gent

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'Dave Gent' text.

June 2013 Ocr Chemistry Past Paper F332:

This book delves into June 2013 Ocr Chemistry Past Paper F332. June 2013 Ocr Chemistry Past Paper F332 is a crucial topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into June 2013 Ocr Chemistry Past Paper F332, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to June 2013 Ocr Chemistry Past Paper F332
- Chapter 2: Essential Elements of June 2013 Ocr Chemistry Past Paper F332
- Chapter 3: June 2013 Ocr Chemistry Past Paper F332 in Everyday Life
- Chapter 4: June 2013 Ocr Chemistry Past Paper F332 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of June 2013 Ocr Chemistry Past Paper F332. The first chapter will explore what June 2013 Ocr Chemistry Past Paper F332 is, why June 2013 Ocr Chemistry Past Paper F332 is vital, and how to effectively learn about June 2013 Ocr Chemistry Past Paper F332.

3. In chapter 2, the author will delve into the foundational concepts of June 2013 Ocr Chemistry Past Paper F332. This chapter will elucidate the essential principles that need to be understood to grasp June 2013 Ocr Chemistry Past Paper F332 in its entirety.

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6. In chapter 5, this book will draw a conclusion about June 2013 Ocr Chemistry Past Paper F332. The final chapter will summarize the key points that have been discussed throughout the book.

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