

MCOLES State Licensing Exam Prep

Questions With Correct Answers 2023/2024

When is the best time to search your patrol vehicle for weapons or contraband?

The beginning of your shift and when non-personnel have been in your vehicle

You can arrest for the following offenses:

- 1) OWI
- 2) Fleeing and Eluding
- 3) Negligent homicide
- 4) No Operator's License
- 5) Reckless Driving
- 6) Hit & Run
- 7) Violation causing death

While on patrol you discover a vehicle parked in a lot behind a jewelry store at 0300 and the motor is running. The proper way to investigate would be to do which of the following?

- (A) Position patrol vehicle at a safe distance and wait for someone to return
- (B) Investigate and see if there is suspicious activity
- (C) Take down license plate number and contact dispatch
- (C) Take down license plate number and contact dispatch

You receive a report from dispatch of a possible domestic dispute in progress. The caller states the neighbors are arguing and requests police. Dispatch says the caller heard gunshots coming from the residence. You arrive and it's quiet. The lights are off. The correct way to respond would be to do which of the following?

- (A) Go to the door, announce your presence, and wait for a response
 - (B) Proceed cautiously and silently enter the house
 - (C) Verify with dispatch that you have the correct address and wait in patrol car to see if there are signs of movement
 - (C) Verify with dispatch that you have the correct address and wait in patrol car to see if there are signs of movement
- The best method of determining a stolen vehicle that is not moving would be to do which of the following?

- (A) Check VIN through LEIN
- (B) Check license plate through LEIN
- (C) Check suspect's driver's license number and plate through LEIN
- (A) Check VIN through LEIN

While on patrol in a dark alley behind a store, you see someone walking towards you. He sees you, reaches in his pocket, puts something in dumpster, then turns and walks away. You should:

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Tobias Bleicker



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web introduction this laboratory exercise relies on a titration technique to determine an unknown concentration of monoprotic acid in solution in the process of titration a basic solution is gradually added to the acidic solution until complete neutralization is obtained

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web with a strong acid and strong base titration strong acids and strong bases refer to species that completely dissociate to form ions in solution by contrast weak acids and bases ionize only partially and the ionization reaction is reversible

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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web goals understand that parts of an acid base titration be able to determine the K_a or K_b from pH data associated with the titration of a weak acid or base be able to determine the molar mass of a solid monoprotic acid from titration data be able to calculate K_{a1} and K_{a2} for a polyprotic acid by the end of this lab students should be able to

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web conclusion this lab evaluated an acid base reaction with the volumetric analysis of a standard solution being naoh reaction with a 2nd substance $\text{H}_2\text{C}_2\text{O}_4$ aq 2NaOH aq $\text{Na}_2\text{C}_2\text{O}_4$ 4h20 to differentiate an acid from a base in a solution a ph scale is used the ph scale ranges from 0 to 14 a ph of 7 is neutral in an acid base reaction

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