

CHEMICAL THERMODYNAMICS

- 1) In open system, system and surroundings exchange.....
 (a) Energy only (b) Matter only
(c) Energy and matter (d) Neither energy nor matter
- 2) An isothermal expansion takes place at.....
 (a) constant pressure (b) constant volume
(c) constant temperature (d) constant temperature & pressure
- 3) In a closed thermos flask few cubes of ice are placed. It is example of.....
 (a) closed system (b) open system
(c) isolated system (d) thermodynamics system
- 4) In an adiabatic compression, the internal energy of the system.....
 (a) decreases **(b) increases** (c) increases or decreases (d) none
- 5) Identify the intensive quantities from the following.....
 (a) volume (b) enthalpy **(c) pressure** (d) none
- 6) Which of the following is extensive property?
 (a) mole (b) mass (c) volume **(d) all are correct**
- 7) The heat change in a chemical reaction at constant pressure is.....
(a) ΔH (b) ΔU (c) ΔT (d) ΔV
- 8) ΔU is always positive when.....
(a) system absorbs heat and work is done on it
 (b) system emits heat and work is done on it
 (c) system emits heat and no work is done on it
 (d) system absorbs heat and work is done by it
- 9) If in a chemical reaction, the products have less energy than the reactants, that reaction is.....
 (a) a reversible reaction **(b) an exothermic reaction**
 (c) an isothermal reaction (d) an endothermic reaction
- 10) A small piece of Na metal is dropped into water. Na bursts into flame, the reaction is:
 (a) neutralisation (b) reversible **(c) exothermic** (d) endothermic
- 11) For the reaction: $C_3H_8(g) + 5O_2(g) \rightarrow 3CO_2(g) + 4H_2O(l)$ at constant temperature, $\Delta H - \Delta U$ is
 (a) $+RT$ **(b) $-3RT$** (c) $+3RT$ (d) $-RT$
- 12) When NH_4Cl is dissolved in water, the solution becomes cold. The change is?
(a) endothermic (b) exothermic (c) super cooling (d) none

Mcqs Of Thermodynamics With Answers

Léo Pomar



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General Questions of Thermodynamics Shivendra Nandan, Prakash Sharma, Satyajeet Kant, Thermodynamics is the branch of physics that deals with the relationships between heat and other forms of energy. In particular, it describes how thermal energy is converted to and from other forms of energy and how it affects matter. *UGC NET Environmental Science 3000 + [MCQ] Question Answer E-book* DIWAKAR EDUCATION HUB, 2022-02-13 UGC NET Environmental Science Unit Wise Practice Question Answer As Per New Updated Syllabus MCQs Highlights 1 Complete Units Cover Include All 10 Units Question Answer 2 300 Practice Question Answer Each Unit 3 Total 3000 Practice Question Answer 4 Try to take all topics MCQ 5 Include Oriented Most Expected Question Answer 6 As Per the New Updated Syllabus Check Sample Pdf Click On Below Link Download Pdf For Free For More Details Call 7310762592 7078549303 [Engineering Physics MCQ \(Multiple Choice Questions\)](#) Arshad Iqbal, The Engineering Physics Multiple Choice Questions MCQ Quiz with Answers PDF Engineering Physics MCQ PDF Download Quiz Questions Chapter 1 36 Practice Tests with Answer Key Physics Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Engineering Physics MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Engineering Physics MCQ PDF book helps to practice test questions from exam prep notes The Engineering Physics MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Engineering Physics Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters Alternating fields and currents astronomical data capacitors and capacitance circuit theory conservation of energy coulomb's law current produced magnetic field electric potential energy equilibrium indeterminate structures finding electric field first law of thermodynamics fluid statics and dynamics friction drag and centripetal force fundamental constants of physics geometric optics inductance kinetic energy longitudinal waves magnetic force models of magnetism newton's law of motion Newtonian gravitation Ohm's law optical diffraction optical interference physics and measurement properties of common elements rotational motion second law of thermodynamics simple harmonic motion special relativity straight line motion transverse waves two and three dimensional motion vector quantities work kinetic energy theorem tests for college and university revision guide Engineering Physics Quiz Questions and Answers PDF free download eBook's sample covers beginner's solved questions textbook's study notes to practice online tests The book Engineering Physics MCQs Chapter 1 36 PDF includes high school question papers to review practice tests for exams Engineering Physics Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET Jobs Entry Level competitive exam Engineering Physics Mock Tests Chapter 1 36 eBook covers problem solving exam tests from physics textbook and practical eBook chapter wise as Chapter 1 Alternating Fields and Currents MCQ Chapter 2 Astronomical Data MCQ Chapter 3 Capacitors and Capacitance MCQ Chapter 4 Circuit Theory MCQ Chapter 5 Conservation of Energy MCQ Chapter 6 Coulomb

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Venkanna,B. V. Swati,2010-07 This book provides an in depth discussion of the principles of thermodynamics It focuses on engineering applications of theory and sound techniques for solving thermodynamic problems The book presents the fundamental concepts of thermodynamics and describes the theory of work and heat The text covers in detail the first law and the second law of thermodynamics with their applications It also explains the concepts of entropy and availability and irreversibility In addition the book presents thermodynamic properties of pure substances ideal gases and mixtures of ideal gases as well as real gases This book is designed for undergraduate students of mechanical engineering industrial and production engineering automobile engineering and aeronautical engineering for their courses in thermodynamics Key Features Presents the text in a simple and elegant manner to enable the students to grasp the essentials of the subject easily and quickly Covers all types of problems of various difficulty levels Includes more than 300 worked out examples and a large number of end of chapter exercises Provides solutions to several model question papers at the end of the book

Thermodynamics Rumi Michael Leigh,2023-05-24 *Thermodynamics Things You Should Know Questions and Answers* explains the principles of thermodynamics in a question and answer format The book introduces energy work heat and the basic laws of thermodynamics It describes the concepts of entropy enthalpy internal energy and equilibrium and shows how these ideas connect to physical and chemical processes Key topics include the first and second laws of thermodynamics the role of entropy in natural systems and the behavior of gases as described by equations of state The text also examines heat transfer efficiency and applications of thermodynamics in engines refrigeration and power systems Worked examples and explanations show how thermodynamic concepts are applied in solving problems and understanding physical phenomena This book will interest students engineering learners and those studying physics who want to understand thermodynamics and its applications *Solutions Manual for Thermodynamics* Jack Philip Holman,1974

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