

# Judicial Branch in a Flash

Name: andy zaruma

**A. Complete the Sentence.** Use the terms and ideas that you learned in this lesson to finish each statement.

- The only court the Constitution creates is Supreme Court
- The two court systems in the United States are state and federal
- Two kinds of legal cases are civil and criminal
- The job of the Court of Appeals is see if trial did something wrong
- It's difficult to take a case to the Supreme Court because they choose what to se and they dont see lots of cases
- If you lose a case in the trial court, you can appeal to a higher court
- If an appellate court affirms a case, it means the verdict stays the same
- If a law is unconstitutional, the Supreme Court can strike it downs
- Evidence is used for prove that someone is inn innocent and to see if the side of story is right
- A trial with no jury is called a bench trial

**B. Making Comparisons.** Decide whether each description fits trial courts only, appellate courts only, or both, and write the letter of the description in the correct part of the diagram. The first one is done for you.



- A. Hears civil cases
- B. Might have a jury trial
- C. Does not hear cases for the first time
- D. Hears criminal cases
- E. Reviews a verdict to look for mistakes
- F. Usually has three-judge panels
- G. Hears cases for the first time
- H. Works with laws

**C. Order in the Court!** Number each set of events to put the three events in the correct order.

- 3 The Court of Appeals remands the case
- 1 A new trial begins
- 2 The first verdict is appealed

- 1 Trial is held in the District Court
- 3 The Supreme Court agrees to hear the case
- 2 An appeal is made to the Court of Appeals

- 2 Evidence is presented to the jury
- 1 The members of the jury are chosen
- 3 The jury returns a verdict

- 3 The Supreme Court strikes down the law
- 2 Supreme Court hears a case about the law
- 1 Congress passes a law

# Judicial Branch In A Flash Answers Icivics

**Michael Seilmaier**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

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web answer  $\text{Na}^+ \text{aq}$  and  $\text{SO}_4^{2-} \text{aq}$  identify the chemical equation the spectator ions  $\text{KOH} \text{aq}$   $\text{HNO}_3$   $\text{KNO}_3 \text{aq}$   $\text{H}_2\text{O}$   $\text{l}$   $\text{K}^+ \text{aq}$   $\text{OH}^- \text{aq}$   $\text{H}^+ \text{aq}$   $\text{NO}_3^- \text{aq}$   $\text{K}^+ \text{aq}$   $\text{NO}_3^- \text{aq}$   $\text{H}_2\text{O}$   $\text{l}$  answer  $\text{K}^+ \text{aq}$  and  $\text{NO}_3^- \text{aq}$  identify the chemical equation the spectator ions  $\text{Pb(NO}_3)_2 \text{aq}$   $2\text{KI} \text{aq}$   $\text{PbI}_2 \text{s}$   $2\text{KNO}_3 \text{aq}$

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web these ions are called spectator ions because they do not participate in the actual reaction canceling the spectator ions gives the net ionic equation which shows only those species that participate in the chemical reaction  $2\text{Ag}^+ \text{aq}$   $\text{CrO}_4^{2-} \text{aq}$   $\text{rightarrow}$   $\text{Ag}_2\text{CrO}_4 \text{s}$  label 4 2 3

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web chemical equation in which all dissolved ionic reactants and products including spectator ions are explicitly represented by formulas for their dissociated ions molecular equation chemical equation in which all reactants and products are

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web jan 29 2020 the ionic form of this reaction is  $2\text{Na}^+ \text{aq}$   $2\text{Cl}^- \text{aq}$   $\text{Cu}^{2+} \text{aq}$   $\text{SO}_4^{2-} \text{aq}$   $2\text{Na}^+ \text{aq}$   $\text{SO}_4^{2-} \text{aq}$   $\text{CuCl}_2 \text{s}$  the sodium ions and sulfate ion are the spectator ions in this reaction they appear unchanged in both the product and reactant side of the equation

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web the six no 3 aq ions and the six na aq ions that appear on both sides of the equation are spectator ions that can be canceled to give the net ionic equation  $3\text{Ba}^{2+} + 2\text{aq} + 2\text{po}_4^{3-} \rightarrow \text{Ba}_3(\text{PO}_4)_2\text{s}$

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web these spectator ions whose presence is required to maintain charge neutrality are neither chemically nor physically changed by the process and so they may be eliminated from the equation to yield a more succinct representation called a

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web video answer you know spectator ions are the ions present on the both sides of the equation indicate cancels out in the total ionic equations so you know in a case chemical reactions are described with chemical equations which symbolically present the starting materials end products

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web the equation below is the final net ionic equation after elimination of the spectator ions which yields water  $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O l}$  the spectator ions in  $\text{KOH aq}$  and  $\text{HNO}_3 \text{ aq}$  are  $\text{K}^+ \text{ aq}$  and  $\text{NO}_3^- \text{ aq}$  how to identify and list the

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web aug 8 2022 a spectator ion is an ion that does not take part in the chemical reaction and is found in solution both before and after the reaction the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction

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web the new format requires students to write balanced chemical equations showing only the reacting substances excluding spectator ions for three different sets of reactants and to answer a short question requiring no calculator about each reaction the questions are intended to test the students understanding of the meaning of the reactions

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web jun 2 2023 the spectator ions are cut out from a reaction hence the net ionic equation for the above reaction would be  $2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} \rightarrow \text{CuCl}_2 \text{s}$   $2\text{Cl}^- \text{aq} + \text{Cu}^{2+} \text{aq} \rightarrow \text{CuCl}_2 \text{s}$  these spectator ions are not actively involved in the reaction the existence of spectator ions affects the Debye length