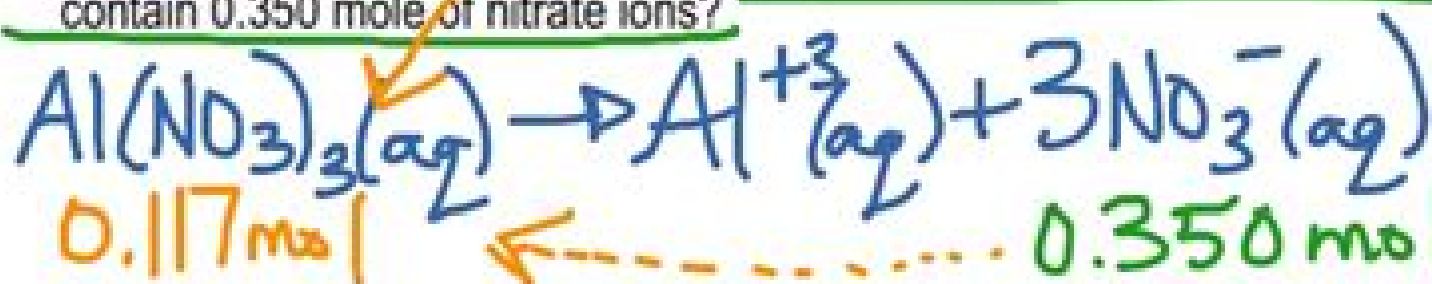
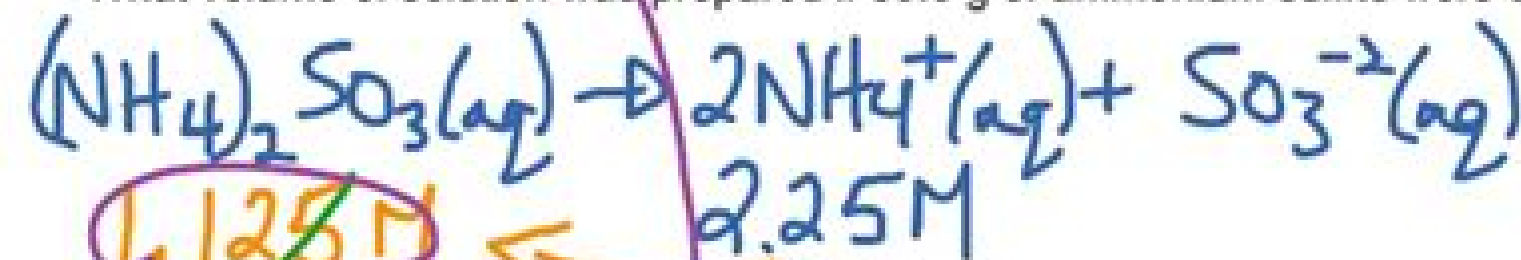


9. A solution of aluminum nitrate has a concentration of 4.5 M. What volume of solution will contain 0.350 mole of nitrate ions?



$$\cancel{0.117 \text{ mol}} \times \frac{1 \text{ L}}{\cancel{4.5 \text{ mol}}} = \underline{0.026 \text{ L}} \text{ or } \underline{26 \text{ mL}}$$

10. A solution of ammonium sulfite is prepared with a concentration of 2.25 M in ammonium ions. What volume of solution was prepared if 50.0 g of ammonium sulfite were used?



$$\cancel{50.0 \text{ g}} \times \frac{\cancel{\text{mol}}}{\cancel{116.18 \text{ g}}} \times \frac{\text{L}}{\cancel{1.12 \text{ mol}}} = \underline{0.384 \text{ L}} \text{ or } \underline{384 \text{ mL}}$$

Molarity Pogil Answer Key

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