

# CHEMICAL EQUILIBRIUM Chapter 9

- (a) Products (b) Reversible reaction  
(c) Reactants (d) Irreversible reaction
- Name the reactants in this equation  $2\text{H}_2 + \text{O}_2 \xrightarrow[\text{Pt}]{\text{heat}} 2\text{H}_2\text{O}$
- (a)  $\text{H}_2\text{O}$  (b)  $\text{H}_2$  and  $\text{O}_2$  (c)  $\text{O}_2$  (d) None of these
- The reaction in which the products do not recombine to form reactants is known as:
- (a) Reversible reactions (b) Reactants  
(c) Products (d) Irreversible reactions

- The reactions in which the products recombine to form reactants are called:
- (a) Reactants (b) Reversible reactions  
(c) Irreversible reactions (d) Both a and b

## QUESTIONS

In chemical reactions the substances that combine are called:

- Irreversible reaction goes on:
- (a) Never complete (b) Both a and c (c) completion (d) None of these
- Reversible reaction:
- (a) can complete (b) never complete (c) Both a and b (d) None of these
7. The colour of hydrogen iodide (HI) is:
- (a) Blue (b) Grey (c) Purple (d) Colourless
8. The colour of iodine ( $\text{I}_2$ ) is:
- (a) Purple (b) Green (c) Yellow (d) None of these
9. At equilibrium the rate of forward and reverse reaction becomes.
- (a) Greater (b) Normal (c) Equal (d) Lesser
10. Dynamic means reaction is:
- (a) Stops (b) Still continuing (c) Opposite direction (d) Both a and b
11. When reaction ceases to proceed it is called:
- (a) Dynamic equilibrium (b) Static equilibrium  
(c) chemical equilibrium (d) none of these
12. Reaction in which reactants react to form products called:
- (a) forward reactions (b) reverse reactions (c) reversible reaction (d) backward reactions
- At initial stage the rate of forward reaction is.
- (a) low (b) very low (c) very fast (d) all of these
14. There are \_\_\_\_\_ possibilities at equilibrium state.
- (a) three (b) five (c) four (d) two
15. Reverse reaction takes place from:
- (a) left to right (b) right to left (c) both a and b (d) none of these
16. Reverse reactions \_\_\_\_\_ gradually.
- (a) speeds up (b) negligible (c) slows down (d) none of above
17. Forward reaction takes place from:
- (a) left to right (b) right to left (c) both a and b (d) all of these

# Mcq On Chemical Equilibrium

**Lei Shi**



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