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- (D) None of these
- The number of tetrahedral and octahedral voids in hexagonal prismatic unit cell HCP is
 - (A) 8,4 (C) 12,6

- (B) = 2,1
- 12,6 (D) 6,12
- 16. Which of the following void is smallest in close packing?
 - (A) Tetrahedral
- (B) Octahedral (D) Square
- (C) Cubic (D)
- The number of octahedral void in BCC structure is
 - (A) 0

(B) 1

(C) 2

- (D) 4
- 18. Which of the following interstitial site is formed when the three closed packed spheres of one layer is put over three closed packed spheres of the second layer, their positions being inverted with respect to each other?
 - (A) Tetrahedral (B) Octahedral (C) Rhombobedral (D) Tetragonal
- 19. In which crystal, alternate tetrahedral voids are occupied?
 - (A) NaCl (C) ZnS

- (B) CaF₂(D) Na₂O
- 20. A metallic element crystallizes into a lattice containing a sequence of layers of ABABAB. Any packing of spheres leaves out voids in the lattice. Approximately what percentage by volume of this lattice is empty space?
 - (A) 74%

(B) 26%

(C) 48%

- (D) 32%
- Packing efficiency of an identical solid sphere is 74% in
 - (A) Simple cubic structure
 - (B) Monoclinic structure
 - (C) HCP structure
 - (D) BCC structure

- 22. Which of the following expression represents total volume of atoms present in a FCC unit cell of a metal? (r is the atomic radius)
 - (A) $\frac{20}{3} \pi r^3$

(B) $\frac{24}{3}\pi r^3$

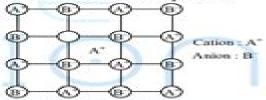
(C) $\frac{12}{3}\pi r^3$

- (D) $\frac{16}{3}\pi r^3$
- In a BCC packing, the nearest neighbours lie along the
 - (A) edges of the cube
 - (B) face diagonal
 - (C) line joining the two opposite corners of the face.
 - (D) cube diagonal
- 24. The ionic radii of Rb* and I are 1.46 and 2.16

A, respectively. The most probable type of structure exhibited by it is

- (A) CsCl type
- (B) NaCl type
- (C) ZnS type
- (D) CaF₂ type
- The radius of Ag* ion is 126 pm while of I* ion is 216 pm. The coordination number of Ag in AgI is
 - (A) 2 (C) 6

- (B) 4 (D) 8
- 26. Structure shown here represents:



- (A) Schottky defect
- (B) Frenkel defect
- (C) Metal excess defect
- (D) None of these
- 27. Select the incorrect statement:
 - (A) Stoichiometry of crystal remains uneffected due to Schottky defect.

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William H. Myers, William T. Mooney

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web apr 4 2018 origami [[[]]] which in japanese literally means folding paper is thought to have originated in japan but is a part of many asian cultures such as chinese through various forms of folding a flat sheet of paper is transformed into different shapes representing a variety of things

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web may 21 2023 while these intricate folded paper creations may appear to be abstract their significance is much deeper than meets the eye there are many symbolic aspects to origami and it is a part of japanese festivals and ceremonies in this article we ll dive into the art of origami and explain why it s so important in japanese culture

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