
Mastering Biology chapter 5

Osmosis: if a semipermeable membrane is placed between the following solutions which will decrease in volume?

A: 1.4% (m/v) starch

B: 7.62% (m/v) starch - answer Solution A

Osmosis: Which solution will increase in volume?

C: 9% (m/v) NaCl

D: 12.4% (m/v) NaCl - answer Solution D

exerts the same osmotic pressure as the cellular fluids/ behaves as if it had the same solute concentration as the cell - answer Isotonic Solution

Behaves as if it had a higher solute concentration than the cell (cause cell to shrink) - answer Hypertonic Solution

Behaves as if it had a lower solute concentration than the cell (cause cell to swell/burst) - answer Hypotonic Solution

A red blood cell placed in a hypertonic solution will shrink in a process called - answer Crenation

A red blood cell in a hypotonic solution will swell and potentially burst in a process called - answer Hemolysis

what makes the tail (composed of fatty acid or isoprene) molecules hydrophobic? - answer the density of carbon-hydrogen bonds, unable to interact with water

The phospholipid head consists of a three-carbon molecule called ___ linked to a phosphate group. The phosphate group is then bonded to another small organic molecule (ex: amino acid serine) - answer glycerol

(true/false) small nonpolar molecules and even small, uncharged polar molecules (water, glycerol) can pass through the selectively permeable cell - answer true

(true/false) polar molecules, such as sugars and ions, which have charges, CANNOT pass through the membrane. - answer true

movement from the area of higher concentration to areas of lower concentration, until the concentrations are equalized throughout the beaker. - answer diffusion

Mastering Biology Answers Chapter 6

David S. Brown



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