Chemistry - Unit 4 Worksheet 4

DMC.5: I can cite and interpret the evidence that supports the belief that some pure substances are made up of simpler particles (atoms) in a definite ratio.

Answer the following questions. Be sure to show all mathematical work and reasoning and use complete sentences in explanations.

- Two compounds of hydrogen and oxygen are tested. Compound I contains 120.0 g of oxygen and 15.0 g of hydrogen. Compound II contains 32.0 g of oxygen and 2.0 g of hydrogen.
 - Determine the ratio of the mass of oxygen to the mass of hydrogen in each of the compounds.
 - b. Why are the compounds not the same?
 - c. What is significant about these mass ratios?
 - d. If compound I is water, what could be the formula of compound II?
- Nitrogen and oxygen combine to form a variety of compounds. The following data were collected for three different compounds of nitrogen and oxygen:

Analysis Data of Nitrogen & Oxygen Compounds	
Compound	Mass of Nitrogen that combines with every 1.0 g of Oxygen
A	1.750 g
В	0.8750 g
C	0.4375 g

- a. Additional evidence shows that the formula of compound B is NO. Sketch particle diagrams of molecules of all three compounds.
- Justify your representations above.
- c. There is no compound between Nitrogen and oxygen with a mass ratio of 2.00 grams of N for every 1.00 grams of oxygen. Why?

Modeling Chemistry Unit 2 Answers

Kathleen Armour

Modeling Chemistry Unit 2 Answers:

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scientists psychometricians and assessment specialists convened to discuss these challenges as part of the Learning Progressions in Science LeaPS conference The LeaPS conference provided a structured forum for considering design decisions entailed in four aspects of work on learning progressions de ning learning progressions developing assessments to elicit student responses relative to learning progressions modeling and interpreting student performance with respect to a learning progressions and using learning progressions to in uence standards curricula and teacher education This book presents speci c examples of learning progression work and syntheses of ideas from these examples and discussions at the LeaPS conference The Educational Review ,1912 Books in Print ,1994 Energy Research Abstracts ,1990 Resources in Education ,1984 **EPA Publications Bibliography Quarterly Abstract Bulletin** United States. Environmental Protection Agency, 1993-07 **Abstracts of Papers Presented to the American Mathematical Society** American Mathematical Society, 2003 Holt Chemistry Ralph Thomas Myers, 2004 **Integrated Mathematics** Holt McDougal,1998 **INIS Atomindeks** .1988 The United States Catalog Mary Burnham, Carol Hurd, 1928 Publishers' Trade List Annual ,1980 Catalog of Copyright Entries. Third Series Library of Congress. Copyright Office.1963 Government Reports Announcements & Index ,1995

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