

Chemistry – Unit 6 Notes 1 - Thomson Model of the Atom

J. J. Thomson performed experiments with cathode rays in an attempt to understand electricity – which was still a mystery in the late 1800s. Use the website *A Look Inside the Atom*¹ to find the conclusions that Thomson and other physicists drew regarding the mysterious cathode rays.

Thomson's 1897 Experiments - state the conclusions Thomson drew from each of his famous cathode ray experiments:

1. **First Experiment:** Thomson directed the beam at an electrometer and tried to separate the evidence of charge from the path of the beam. *What connection did Thomson find between charge and the cathode rays? Was the charge positive or negative?*
2. **Second Experiment:** Thomson tried passing the cathode ray through an electric field. *How did cathode ray beam behave when it passed through an electric field? What did he conclude after his second experiment?*
3. **Third Experiment:** Thomson did some careful measurements on how much the path of the cathode ray was bent in a magnetic field and how much energy they carried. From this work Thomson could describe the mass/charge ratio of the cathode ray particles. *What amazing result did Thomson find?*

Thomson's Atomic Model: Thomson presented three hypotheses from his experiments. Only two were accepted by physicists – in fact the third was shown to be wrong! From the first two came a model of the atom known as the *Plum Pudding* model. Complete the atom drawing below to illustrate Thomson's plum pudding model. Explain how this fits with his observations.



¹ <https://www.aip.org/history/exhibits/electron/jjrays.htm>
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Whispering the Secrets of Language: An Psychological Journey through **Modeling Chemistry U6 Test Key**

In a digitally-driven earth wherever monitors reign supreme and instant communication drowns out the subtleties of language, the profound secrets and psychological subtleties concealed within phrases usually go unheard. Yet, located within the pages of **Modeling Chemistry U6 Test Key** a charming literary value blinking with fresh thoughts, lies a fantastic quest waiting to be undertaken. Published by a skilled wordsmith, this charming opus invites readers on an introspective journey, softly unraveling the veiled truths and profound affect resonating within the very cloth of each word. Within the psychological depths of this touching evaluation, we can embark upon a genuine exploration of the book is key themes, dissect their captivating writing type, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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