Classify each reaction as synthesis (S), decomposition (DR), single-replacement (SR), double-replacement (DR), or combustion (C). Then predict the products of the products of the combustion (C). Then predict the products of the combustion and write the balanced equation.	e #
reaction and write the con- reaction and the con- reaction and the con- reaction and	
Lead + zinc acetate →	
Calcium + oxygen → Barium carbonate →	
Methane (CH, problems products of che silver nitrate (+ identifying r	for prediction react eaction type
Culfuric acid →	

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