

Electron Configuration Practice Worksheet **KEY**

In the space below, write the unabbreviated electron configurations of the following elements:

- 1) sodium $1s^2 2s^2 2p^6 3s^1$
- 2) iron $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^6$
- 3) bromine $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^5$
- 4) barium $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2$
- 5) neptunium $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^6 7s^2 5f^3$

In the space below, write the abbreviated electron configurations of the following elements:

- 6) cobalt $[Ar] 4s^2 3d^7$
- 7) silver $[Kr] 5s^2 4d^9$
- 8) tellurium $[Kr] 5s^2 4d^{10} 5p^4$
- 9) radium $[Rn] 7s^2$
- 10) lawrencium $[Rn] 7s^2 5f^{14} 6d^1$

Determine what elements are denoted by the following electron configurations:

- 11) $1s^2 2s^2 2p^6 3s^2 3p^4$ **Sulfur**
- 12) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^1$ **Rubidium**
- 13) $[Kr] 5s^2 4d^{10} 5p^3$ **Antimony**
- 14) $[Xe] 6s^2 4f^{14} 5d^6$ **Osmium**
- 15) $[Rn] 7s^2 5f^{11}$ **Einsteinium**

Determine which of the following electron configurations are not valid:

- 16) $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 4d^{10} 4p^5$ **should be $3d^{10}$**
- 17) $1s^2 2s^2 2p^6 3s^2 3d^5$ **$3s^2$ is impossible & "p" instead of "d"**
- 18) $[Ra] 7s^2 5f^8$ **Ra is not a noble gas**
- 19) $[Kr] 5s^2 4d^{10} 5p^5$ **Valid**
- 20) $[Xe]$ **Element can't be its own electron configuration.**

Ion Electron Configuration Worksheet 2014

Michael Brown

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Ion Electron Configuration Worksheet 2014:

The book delves into Ion Electron Configuration Worksheet 2014. Ion Electron Configuration Worksheet 2014 is an essential topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Ion Electron Configuration Worksheet 2014, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Ion Electron Configuration Worksheet 2014
 - Chapter 2: Essential Elements of Ion Electron Configuration Worksheet 2014
 - Chapter 3: Ion Electron Configuration Worksheet 2014 in Everyday Life
 - Chapter 4: Ion Electron Configuration Worksheet 2014 in Specific Contexts
 - Chapter 5: Conclusion
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 6. In chapter 5, the author will draw a conclusion about Ion Electron Configuration Worksheet 2014. This chapter will summarize the key points that have been discussed throughout the book.
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