#### Reminders:

 In a neutral atom the number of protons equals the number of electrons.

An atom can
 NEVER gain or lose
 protons

 The number of protons equals the atomic number

#### NAME:

# Ion Practice Set

- 1. What is an ion?
- 2. What does the number next to the ions signify?

#### Complete the following table, using the periodic table in the back of your book.

	ELEMENT NAME	ION SYMBOL	NUMBER OF PROTONS	NUMBER OF ELECTRONS	NUMBER OF ELECTRONS LOST OR GAINED
ех	Fluorine	F <sup>-</sup>	9	10	gained one
1			53	54	
2			16		gained two
3	potassium				lost one
4		Ca <sup>+2</sup>			
5			35	36	
6		Sr <sup>+2</sup>			
7		н*			
8		_	8		gained two
9			12		lost two
10	aluminum			10	
11			34	36	
12		H <sup>-</sup>			
13	lithium				lost one
14		Rb*			
15			17	18	

# **Ion Practice Set Key**

**M Woodhall** 

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The Balance Sheet, 1929

Educart CBSE Class 12 Chemistry One Shot Question Bank 2026 (Includes PYQs for 2025-26) , Our Navy, the Standard Publication of the U.S. Navy ,1918 The Balance Sheet , American Shorthand Teacher ,1950 A Course in Radio Fundamentals George Grammer,1942 The Business Education World John Robert Gregg,1950 New MCAT 45 2007 Kaplan,2006 Complete test information and essential test taking strategies Concrete advice about reading and handling the most difficult physical science biological science and verbal reasoning passages High level challenging practice sets the toughest questions for each section of the MCAT Writing sample advice benchmark essay examples and scoring guidelines American Accountant ,1928 The Radio Amateur's Handbook ,1942 Overland Monthly ,1884

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