Start Quiz 2 Material

Leaving Group Ability

- Weakest base is best, most stable leaving group
- Find the weakest base because conjugate acid has lowest pKa

How Carboxylic Acids and Derivatives React

Nomenclature

Acyl chloride (acid chloride)



Carboxylic acid



Ester



Benzoic Acid



Acid

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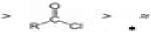


Lon

L_{NH2}

Amide.

Relative Reactivities of Carboxylic Acids and Derivatives











chloride > acid anhydride > carboxylic acid = ester > amide

 Resonance argument explains reactivities of carboxylic acids/esters versus amides because resonates to 0 or N with + FC. N is less EN than 0 so more stable with +FC. Carboxylic acids and esters are therefore more susceptible to nucleophilic addition.

Acid Chlorides

- Reactant: acyl chloride
- Reagent: nucleophile (alcohol, water, ammonia, etc)
- Product: depends on nucleophile (ester, carboxylic acid, amide)
- Mechanism:

- So reactive that they react under neutral conditions
- No S₂2 at sp² so always write the tetrahedral intermediate
- Sequence called (1) addition-elimination or (2) nucleophilic acyl substitution

National Standardized Organic Chemistry Test Study Guide

Shasha Hu

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