



2023 Key Stage 2 Mathematics Mark Scheme - All on One A4 Sheet



* There is additional guidance in the mark scheme including guidance on answers that may be worthy of credit if the answer is incorrect.
Each question is worth 1 mark unless otherwise stated (e.g. Q26g indicates 2 marks). Full mark schemes can be found [here](#).

Paper 1 Arithmetic

Qn.	Answer
1	697
2	6,594
3	2,808
4	8,413
5	240
6	960
7	14,753
8	2,754
9	50
10	520
11	400
12	6
13	900
14*	10/63
15	83
16*	13/16
17	0.69
18*	17/18
19	13,175
20*	37,592 (2m)
21*	1/24
22*	2
23	78
24	38.4
25*	1.3 (2m)
	1,349 r1 OR
26*	1,349 r2 OR
	1,349 1/5
27*	364
28*	1/18
29*	224,763 (2m)
30*	171
31	14.8
32*	3/10
33*	1.72 (2m)
34*	2 1/12 OR
	25/12
35*	285
36*	600

Paper 2 Reasoning

Qn.	Answer
1	
2*	
3	(5, 2)
4	7
	A correct explanation, e.g.
	• Uses 3 cars and 3 motorcycles to show that number of wheels cannot be 28 because there are 26 wheels.
	• Uses 3 motorcycles and the sum of 28 wheels to show that the number of cars cannot be 3.
	• Uses 3 cars and the total number of 28 wheels to show that the number of motorcycles cannot be 3.
	• Demonstrates that Stefan would have either two more wheels or an extra motorcycle.
5*	23.61
7*	75 100 250
8*	24,800
	All three diagrams matched correctly
9*	
10	216
11a	20,519
11b	18,939
12*	
13	110
14	Answer in the range of 138 to 152 inclusive
15*	
16*	5
17*	13 (2m)
18*	3/20 (2m)
19	200
20*	
21*	267.5 OR 267 1/2 (m) (2m)
22*	3/52
23*	19 (2m)
24	564
25a	20
25b*	4.8 kg (2m)
26a	18
26b	3

Paper 3 Reasoning

Qn.	Answer
1	
2*	
3*	
4	1,700 1,800 1,900 2,000 2,100
5*	13.2 14.7 15.3 16.3 17.8
6	300,000
7*	10 5/12
8*	Shapes matched as shown: Explanation that recognises that 32 is not a multiple of 3 OR a description that includes one or both of the multiples of 3 either side of 32
10*	3
11*	1,458 (2m)
	Two marks for two boxes correctly ticked, and no incorrect boxes ticked
12*	ONE mark for two boxes ticked correctly and one incorrect box ticked OR only one box ticked correctly and no incorrect boxes ticked
13	9
14a*	1/4
14b*	1/5
15a*	1:50 OR 09:50 pm OR 17:50
15b*	1 (hours) 45 (minutes)
16*	350g (2m)
17*	5/6
18a*	Drawing in the range 650mm to 750mm
18b*	1,343 (2m)
19*	102.65 (2m)
	Two marks for three correct expressions circled. One mark for two correct expressions circled and none incorrect
21*	32.3 (3m)
22	range
	Two marks for both correct.
23	One mark for one correct.

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J Spring



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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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oct 2 2019 the net ionic equation is a chemical equation for a reaction that lists only those species participating in the reaction the net ionic equation is commonly used in acid base

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jun 23 2022 what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic

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aug 10 2023 write the full ionic and net ionic equations for this reaction solution the elemental metals magnesium on the reactant side copper on the product side are neutral

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 $2\text{Ag}^+ \text{aq} + 2\text{F}^- \text{aq} + 2\text{NH}_4^+ \text{aq} + \text{Cr}_2\text{O}_7^{2-} \text{aq} \rightarrow \text{Ag}_2\text{Cr}_2\text{O}_7 \text{s} + 2\text{NH}_4^+ \text{aq} + 2\text{F}^- \text{aq}$ because two NH_4^+ aq

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