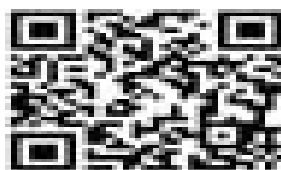


Labpaq Scientific Method Essay



The Scientific Method
Portland Community College Staff* Version 42 0207 00 01

Lab Report Assistant

This document is not meant to be a substitute for a formal laboratory report. The Lab Report Assistant is simply a summary of the experiment's questions, diagrams if needed, and data tables that should be addressed in a formal lab report. The intent is to facilitate students writing of lab reports by providing this information in an editable file which can be sent to an instructor.

observations

Data Table 1: Making Observations | Procedure ... Show more content on Helpwriting.net ...
The pieces were clearer and sharper looking than the original rock salt.

2. Describe the initial observations made when several drops of water were added to few particles rock salt and the observations made several minutes later. Initially, the rock salt partially dissolved in the drops of water. After several minutes, the salt was more dissolved, but had not completely dissolved. There were very fine grains of salt that were visible at the base of the water bubble that hadn't dissolved.

3. In light of these observations, what can be concluded about salt?

Salt can be broken down and dissolved by water, but not in high concentrations.

C. Do all the grains of sand look like they are made out of the same material? No.

1. Propose a hypothesis to explain the observed appearance of sand.

Sand is a conglomeration of rocks and stones that have been broken down into sediments.

2. Describe what was initially observed when several drops of water were added to several grains of sand and the observations made several minutes later.

Initially, when water was added to sand, the sand was unchanged. After several minutes, the sand continued to appear unchanged.

3. In light of these observations, what can be concluded about sand? Sand cannot be dissolved into water.

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Edited by: Kisak



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Introduction to Laboratory and Scientific Methodology Gaston M. U. Pfluegl, 2019 *Laboratory Manual for Principles of General Chemistry* J. A. Beran, Mark Lassiter, 2022-08-09

Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program. The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure, selecting the proper apparatus, employing the proper techniques, systematically analyzing and interpreting the data, and minimizing inherent variables. As a result of good data, a scientific and analytical conclusion is made, which may or may not be right but is certainly consistent with the data. Experiments write textbooks; textbooks don't write experiments. A student's scientific literacy grows when experiences and observations associated with the scientific method are encountered. Further experimentation provides additional cause-effect observations, leading to an even better understanding of the experiment. The 11th edition's experiments are informative and challenging while offering a solid foundation for technique, safety, and experimental procedure. The reporting and analysis of the data and the pre and post lab questions focus on the intuitiveness of the experiment. The experiments may accompany any general chemistry textbook and are compiled at the beginning of each curricular unit. An Additional Notes column is included in each experiment's Report Sheet to provide a space for recording observations and data during the experiment. Continued emphasis on handling data is supported by the Data Analysis section.

Life Sciences 23L Gaston M. U. Pfluegl, 2014 **A Beginner's Guide to Scientific Method** Stephen Sayers Carey, 1994. This is a concise introductory book that provides an introduction to the sciences and scientific method. It is comprehensive, readable, non-technical, and challenging in its ideas. A BEGINNERS GUIDE TO SCIENTIFIC METHOD requires students to use the scientific method to design experiments and assess their own results. The book is brief enough to be used as a supplementary text in a first course in any of the sciences, though comprehensive enough to be used as the primary text in a course in scientific reasoning.

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