

MATH HANDBOOK TRANSPARENCY MASTER

1

Scientific Notation

Use with Appendix B,
Scientific Notation

Scientists need to express small measurements, such as the mass of the proton at the center of a hydrogen atom (0.000 000 000 000 000 000 000 001 673 kg), and large measurements, such as the temperature at the center of the Sun (15 000 000 K). To do this conveniently, they express the numerical values of small and large measurements in scientific notation, which has two parts.

A number in which only one digit is placed to the left of the decimal

$$\longrightarrow N \times 10^n \longleftarrow$$

An exponent of 10 by which the number is multiplied

Thus, the temperature of the Sun, 15 million kelvins, is written as 1.5×10^7 K in scientific notation.

Positive Exponents Express 1234.56 in scientific notation.

Each time the decimal place is moved one place to the left,

$$1234.56$$

$$1234.56 \times 10^0 = 123.456 \times 10^1$$

$$123.456 \times 10^1 = 12.3456 \times 10^2$$

$$12.3456 \times 10^2 = 1.234\,56 \times 10^3$$

$$1.234\,56 \times 10^3$$

the exponent is increased by one.

Negative Exponents Express 0.006 57 in scientific notation.

Each time the decimal place is moved one place to the right,

$$0.006\,57$$

$$0.006\,57 \times 10^0 = 0.0657 \times 10^{-1}$$

$$0.0657 \times 10^{-1} = 0.657 \times 10^{-2}$$

$$0.657 \times 10^{-2} = 6.57 \times 10^{-3}$$

$$6.57 \times 10^{-3}$$

the exponent is decreased by one.

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M Tight



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