

# Ionic Reactions

## Lab Report (Week 2)

Name: \_\_\_\_\_

### Questions

Compare your results with the solubility rules and/or solubility table in your chemistry text. Do your results agree with your expectations from the solubility rules/table?

**Sodium Phosphate ( $\text{Na}_3\text{PO}_4$ ).** Sodium Phosphate did not have a visible reaction with any of the chemicals tested in this lab.

Results agree with the results of the solubility rules.

**Sodium Sulfide ( $\text{Na}_2\text{S}$ ).** Sodium Sulfide was soluble with Copper Sulfate ( $\text{CuSO}_4$ ), where insoluble  $\text{S}$  produced by color change. Sodium Sulfide did not have a visible reaction with any other chemicals in this lab.

Results agree with the results of the solubility rules.

**Sodium Sulfate ( $\text{Na}_2\text{SO}_4$ ).** Sodium Sulfate was soluble with all chemicals in this experiment, the exception of Barium ( $\text{Ba}^{2+}$ ). When combined with Sulfuric Acid was a color change from that was seen in visible reaction.

Results agree with the results of the solubility rules.

**Sodium Chloride ( $\text{NaCl}$ ).** Sodium Chloride easily soluble with the rest of the chemicals in this lab.

Results agree with the results of the solubility rules.

# Labpaq Ionic Reactions Report Answers

**DP Hallahan**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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