

# International Kangaroo Mathematics Contest 2010

Benjamin Level: Class (5 & 6)

Max Time: 2 Hours

## 3-point problems

**Q1)** Knowing that  $\blacktriangle + \blacktriangle + 6 = \blacktriangle + \blacktriangle + \blacktriangle + \blacktriangle$ , which number is hidden by  $\blacktriangle$ ?

- A) 2      B) 3      C) 4      D) 6

**Q2)** The number 4 is next to two mirrors so it reflects twice as shown. When the same thing happens to number 5, what do we get instead for the question mark?



- A) 2      B) 2      C) 5      D) 5

**Q3)** Kalim goes directly from Zoo to School. He counts each flower on the way. Which of the following number can not be his result.



- A) 9      B) 10      C) 11      D) 12

**Q4)** A ladder has 21 stairs. Nadeem and Mahmood are counting stairs; one – from bottom to top, another – from top to bottom. They met on a stair that was called the 10<sup>th</sup> by Nadeem. What number will Mahmood give to this stair?

- A) 13      B) 11      C) 12      D) 10

**Q5)** Adil has connected all the upper points to all the lower points. How many lines Adil has drawn?



- A) 20      B) 25      C) 30      D) 35

**Q6)** A fly has 6 legs, while a spider has 8 legs. Together, 2 flies and 3 spiders have as many legs as 10 birds and

- A) 3 cats      B) 4 cats      C) 5 cats      D) 6 cats

**Q7)** There are seven bars in the box. It is possible to slide the bars in the box so there will be space for one more bar. At least how many bars have to be moved?



- A) 1      B) 2      C) 3      D) 4

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**JG Myers**



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1s22s22p63s23p64s23d104p65s24d6 ruthenium 23 kr 5s24d10 cadmium 24 xe 6s24f145d106p2 lead 25 rn 7s25f146d4 seaborgium determine if the following electron configurations are correct 26 1s22s22p63s23p64s24d104p65s1 no it should be 3d10

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web how many protons neutrons and electrons are in atoms of these isotopes write the complete electron configuration for



each isotope answer co has 27 protons 27 electrons and 33 neutrons  $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^7$  i has 53 protons 53 electrons and 78 neutrons  $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^5$

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web select the electron configuration tab on show number of electrons introduction beyond argon it is a bit tricky to determine which subshell gets filled next there are several rules that scientists use to determine the electron configurations of larger atoms question how are the electron configurations of elements beyond argon determined