

ACS Review: Practice Exam 1 2025 Final Questions and answers

Identify the true statement:

The atomic number of an element is the number of protons in one atom.

The mass number of an atom is the number of protons in the nucleus plus the number of electrons outside the nucleus.

The nucleus of an atom contains neutrons and electrons.

The number of electrons outside the nucleus is the same as the number of neutrons in the nucleus. - **ANS-**The atomic number of an element is the number of protons in one atom.

What is the valence electron configuration for the element in period 4, group 5A? - **ANS-**
 $4s^2 4p^3$

An atom of Zn has two 4s electrons and ten 3d electrons. How many unpaired electrons are there in the Zn^{2+} ion? - **ANS-**0

Which term best describes the relationship between hydrogen and deuterium?

Deuterium and hydrogen are isotopes.

Deuterium is a polymer of hydrogen.

Deuterium and hydrogen are allotropes.

Deuterium and hydrogen are isomers. - **ANS-**Deuterium and hydrogen are isotopes.

Determine the species that is isoelectronic with the noble gas Kr. - **ANS-** Se^{2-}

From which of the following can the presence of discrete (quantized) energy levels in an atom be inferred?

atomic line spectra

X-ray diffraction by crystals

studies of the photoelectric effect

electron diffraction by crystals - **ANS-**atomic line spectra

Consider the following compound.

Which of the following compounds is a structural isomer of the given compound? - **ANS-**

The generic molecule AB_2 contains two single bonds and one lone pair of electrons around the central A atom.

What is the expected molecular geometry of this molecule? - **ANS-**bent

What is the molecular geometry of IF_3 ? - **ANS-**t-shaped

How many total valence electrons are in the CH_2Cl_2 Lewis structure? - **ANS-**20

Select the species that has both ionic and covalent bonds. - **ANS-** $Al(OH)_3$

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