

SEPARATING MIXTURES

	Required materials Method name: Density <ul style="list-style-type: none"> Objects of different densities Water 	Describe what happens Place objects of different densities into water. Denser objects sink will follow objects float
	Method name: Magnet <ul style="list-style-type: none"> Metal (iron) Nonmetal items Magnet 	
	Method name: Filter <ul style="list-style-type: none"> Small particles Liquid Filter 	
	Method name: Sieve <ul style="list-style-type: none"> Small particles Large particles Sieve 	

Name: _____

MIXTURES & SOLUTIONS Study Guide

Rate each of the following skills:

1 = Very difficult for me 2 = Need some help
3 = I can almost do this 4 = I can teach a friend

..... I can distinguish between a mixture and solution.
 I can describe materials that can dissolve in water.
 I can explain how to separate mixtures and which method to use.

Word	Definition
Mixtures	A heterogeneous combination of substances that keep their identity
Solution	Mixture created when one substance dissolves in another
Dissolve	Breaks down and becomes part of a liquid.
Solvent	The liquid that dissolves the substance.

Highlight any phrases that are describing a solution.

- Includes only solid substances
- Includes 2 liquids
- Includes a liquid and a solid
- Most can be separated
- Most cannot be separated
- Something dissolves

Circle the mixtures below

ANSWER KEY INCLUDED

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to dissolve quicker?

- Increase the temperature of the solvent
- Stir the substance in the solvent
- Cut the solute into smaller pieces- this creates more surface area

Salt
Sugar
Coffee/cocoa

Name 3 substances that do NOT dissolve in water.

Sand
Pepper
Oil



What can you expect to happen when you mix oil and water?

The oil may look like it dissolves at first, but it will separate from the water and either sink or float.

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Mixtures And Solutions Study Guide

Karen C. Timberlake



Mixtures And Solutions Study Guide:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03

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Student's Study Guide & Solutions Manual John Krenos, Joseph Potenza, Loretta Jones, Lynn Koplitz, Thomas

Spence, 2004-03-19 This combination manual is designed to help students avoid common mistakes and understand the material better. The solutions manual section includes detailed answers and explanations to the odd-numbered exercises in the text. Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition

John McMurry, Susan McMurry, 1999 Provides worked-out solutions to text problems along with chapter-by-chapter outlines and a variety of self-tests at the end of each chapter. **Study Guide/Selected Solutions Manual** Julie R.

Frentrup, 2002-06 Contains a brief overview of every chapter, review of skills, self-tests, and the answers and detailed solutions to all odd-numbered end-of-chapter problems in the text book. **Basic Concepts of Chemistry, Study Guide** Leo J.

Malone, 2000 The text's three main goals are to introduce chemistry as a living, relevant science; to encourage learning and critical thinking; and to help readers overcome the math difficulties that impede their progress in chemistry. Designed to help readers master the principles of general chemistry, as a prep book it promotes active involvement with the material. There are special features throughout that reinforce concepts and help to develop strong problem-solving and study skills. Updated to include an Interactive LearningWare problems CD containing several of the chapter-ending problems from the book in an

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