

Name: \_\_\_\_\_

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## Mixed Mole Problems Worksheet

Directions: Solve the following problems. Be sure to show all your work.

1. Find the number of moles if there are  $7.63 \times 10^{24}$  molecules of a substance.
2. How many moles of  $\text{CaCO}_3$  are there in 2.00 kg of  $\text{CaCO}_3$ ?
3. What is the mass of 8.42 moles of  $(\text{NH}_4)_2\text{S}$ ?
4. Find the number of carbon atoms in 18.5 g of  $\text{CH}_3\text{OH}$  (methanol).
5. If one litre of a gas at STP has a mass of 3.17 g, what is the molar mass of this gas?  
Name the diatomic element that makes up this gas?
6. What volume at STP would 11.50 g of oxygen gas occupy?
7. What is the mass of 33.2 L of  $\text{SO}_2$  gas at STP?
8. Calculate the molar mass of a gas at STP, if 360.0 mL of this gas has a mass of 0.680 g

# Mixed Mole Problems Show Work

**L Darling-Hammond**



## Mixed Mole Problems Show Work:

**Chemistry** Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

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**Chemistry Homework** Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more

**Teaching Science for Understanding** James Joseph Gallagher, 2007 Offers middle and high school science teachers practical advice on how they can teach their students key concepts while building their understanding of the subject through various levels of learning activities

**The Farmer's Guide to Agricultural Research in ...** Royal agricultural society of England, London, 1926

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**Material Balances for Chemical Reacting Systems** R.L. Cerro, B.G. Higgins, S. Whitaker, 2022-12-05 Written for use in the first course of a typical chemical engineering program Material Balances for Chemical Reacting Systems introduces and teaches students a rigorous approach to solving the types of macroscopic balance problems they will encounter as chemical engineers This first course is generally taken after students have completed their studies of calculus and vector analysis and these subjects are employed throughout this text Since courses on ordinary differential equations and linear algebra are often taken simultaneously with the first chemical engineering course these subjects are introduced as needed Teaches readers the fundamental concepts associated with macroscopic balance analysis of multicomponent reacting systems Offers a novel and scientifically correct approach to handling chemical reactions Includes an introductory approach to chemical kinetics Features many worked out problems beginning with those that can be

solved by hand and ending with those that benefit from the use of computer software This textbook is aimed at undergraduate chemical engineering students but can be used as a reference for graduate students and professional chemical engineers as well as readers from environmental engineering and bioengineering The text features a solutions manual with detailed solutions for all problems as well as PowerPoint lecture slides available to adopting professors

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