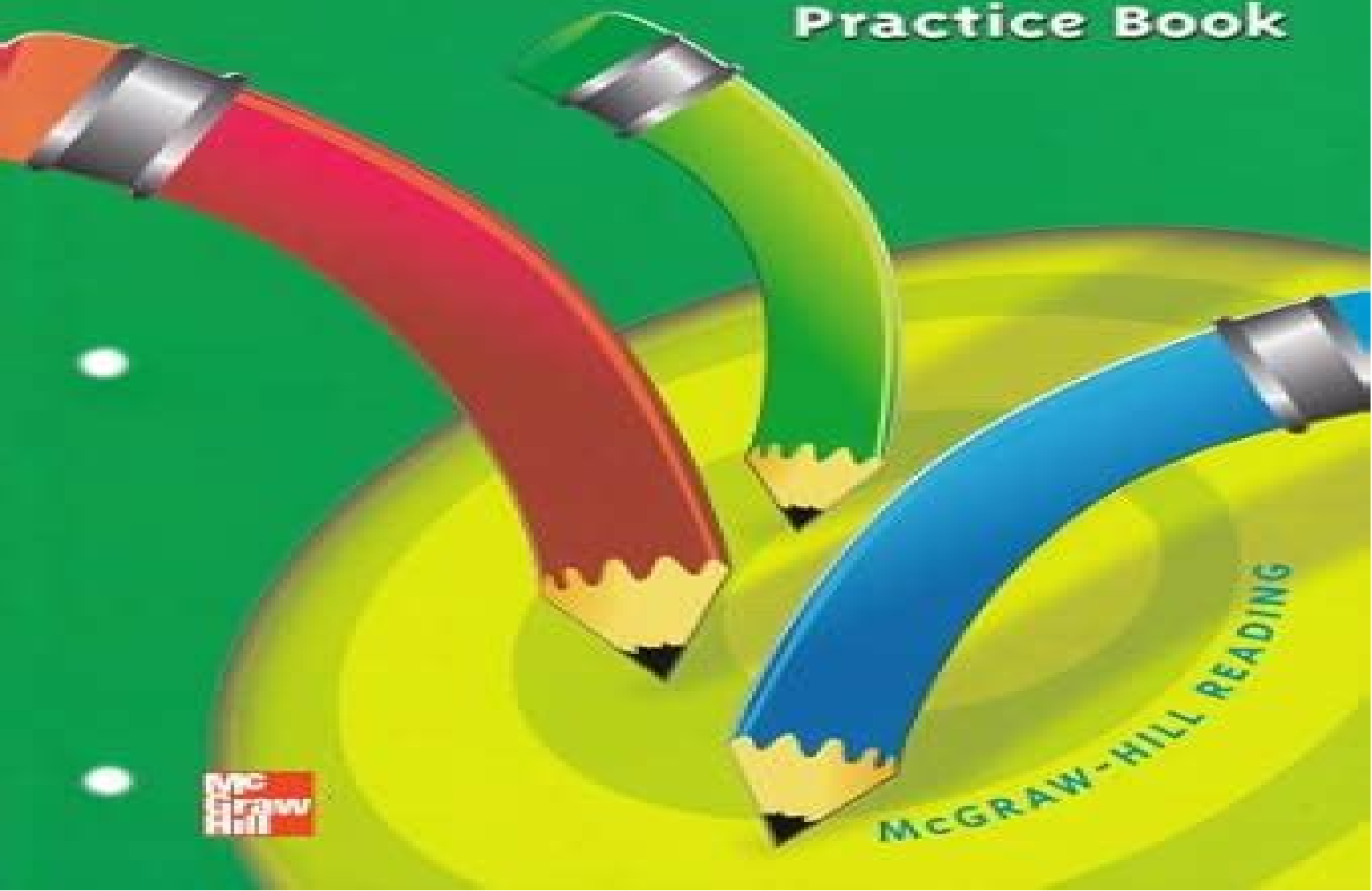


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web solution verified by toppr isotopes same atomic number but different mass number example ${}^{12}_6\text{C}$ ${}^{14}_6\text{C}$ isobars isobars are atoms of different elements having same mass number these have equal number of nucleons but different number of protons neutrons and electrons was this answer helpful 0 0 similar questions define isobars and