

06. The unit of  $K_c$  for reaction  $N_2 + O_2 \rightleftharpoons 2NO$

- (a)  $\text{mol dm}^{-3}$   
(c)  $\text{mol}^{-1} \text{dm}^3$

- (b)  $\text{mol}^{-2} \text{dm}^6$   
(d) ☒ no unit

07. The system is stable in equilibrium when:

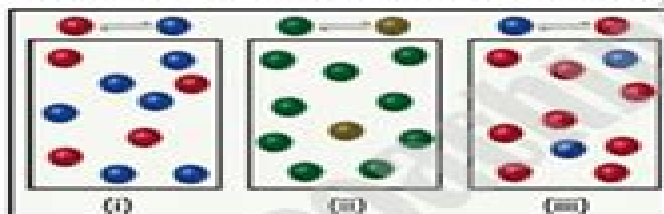
- (a) ☒  $Q_c = K_c$   
(c)  $Q_c < K$

- (b)  $Q_c > K_c$   
(d) None of these

08.  $Q_c$  can be defined as:

- (a) ratio of product and reactants  
(b) ☒ ratio of molar concentration of product and reactant at specific time  
(c) ratio of molar concentration of product and molar volume of reactant  
(d) ratio of molar concentration of product and reactant raised to the power of coefficient

09. Which of the following represent backward reaction?



- (a) (i) and (ii)  
(c) (ii) only

- (b) (ii) and (iii)  
(d) ☒ (iii) only

10. The value of  $K_c$  increases when:

- (a) [Product] is less  
(c) [Reactant] is more

- (b) ☒ [Product] is more  
(d) [Reactant = product]

11. Which of these are reversible reactions?

- (a) Condensation  
(c) ☒ Condensation & Evaporation

- (b) Evaporation  
(d) None of them

# Multiple Choice Questions On Chemical Equilibrium

**Mina Lebitz**



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