

Question Number	Acceptable Answers	Reject	Mark
4 (f) (i)	$(7.5 \times 0.81) = 6.075 / 6.08 \text{ (g)}$ Ignore sf except 1 sf	6.07 Wrong units	1

Question Number	Acceptable Answers	Reject	Mark
4 (f) (ii)	<p><b>Look at final answer.</b>  <b>67% scores 3 marks; answers with 3sf rounding to 67 score 2 marks.</b>          If this is incorrect follow this scheme:</p> <p><b>METHOD 1</b></p> <p>Mol butan-1-ol = <math>(6.075/74)</math>  <math>= 0.0820945</math> (1)</p> <p>maximum mass 1-bromobutane =  <math>(0.0820945 \times 137) = 11.246959 \text{ g}</math> (1)</p> <p>% yield = <math>((7.5/11.24659) \times 100)</math>  <math>= 66.85)</math>  <math>= 67\% \text{ to 2 sf}</math> (1)</p> <p><b>OR METHOD 2</b></p> <p><math>7.5/137 = 0.0547445 \text{ mol (bromobutane)}</math>          (1)</p> <p><math>6.075/74 = 0.0820945 \text{ mol butan-1-ol}</math>          (1)</p> <p>% yield =  <math>((0.05474455) \times 100 / 0.0820945)</math>  <math>= 66.85)</math>  <math>= 67\% \text{ to 2 sf}</math> (1)</p> <p>Also TE from one step of the calculation to the next and TE on 4f(i) unless yield &gt; 100%.</p> <p>Use of 6.08 gives 0.082161 mol,          11.256216 g bromobutane, final answer 67%</p> <p>11.3g bromobutane gives 66%.</p>	<p>Percentages calculated from volumes with no conversion to mol or mass.</p> <p><math>6.075/7.5 \times 100 = 81\%</math> scores 0</p> <p>67.0 (This is 3sf)</p>	3

**Total for Question 4 = 15 marks**

# January 2015 Edexcel Sa Level Xtremepapers

**LP Steffe**



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