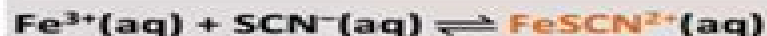


## IA lab work on spectrophotometric determination of $K_c$ of the Iron(3)Thiocyanate system



$$K = \frac{[\text{FeSCN}^{2+}]}{[\text{Fe}^{3+}][\text{SCN}^{-}]}$$

Since I will not change the temperature in my experiment, the equilibrium constant will stay the same, and therefore I can measure the light absorbance and use the Lambert Beer-Law to find the initial concentration of each species in the system and then calculate the  $K_c$ .

$$X(\text{initial concentration of Iron3 and Thiocyanate}) = A(\text{absorbance of Iron(3)Thiocyanate solution}) / \epsilon \cdot l$$

### Equipment:

- 4 test tubes with 4 corks
- test tube rack
- 10 ml pipette with pump thing

### Solutions:

- 20 ml of Iron (3) solution
- 20 ml of Thiocyanate solution
- 20 ml of water  $\text{H}_2\text{O}$

Samples	The amount of Thiocyanate solution $\text{SCN}^{-}$	The amount of water $\text{H}_2\text{O}$	Iron 3 solution $\text{Fe}^{3+}$
1	2 ml	8 ml	5 ml
2	4 ml	6 ml	5 ml
3	5 ml	5 ml	5 ml
4	9 ml	1 ml	5 ml

# Kc Iron Thiocyanate Lab Report

**AW Rasmussen**



## **Kc Iron Thiocyanate Lab Report:**

## Decoding **Kc Iron Thiocyanate Lab Report**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Kc Iron Thiocyanate Lab Report**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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