



Manual For Acer Aspiree5

Puja Mehta



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 $\text{O} + \text{O}_2 \rightarrow \text{O}_3$ slow
 a write the overall balanced chemical equation
 $2\text{O}_3 \rightarrow 3\text{O}_2$
 b identify any intermediates within the mechanism
 c what is the order with respect

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 $k_1 \text{A} + 2\text{B} \rightarrow \text{C}$
 consider the following mechanism
 $\text{O}_3 \rightarrow \text{O}_2 + \text{O}$ fast
 $\text{O} + \text{O}_2 \rightarrow \text{O}_3$ slow
 a write the overall balanced chemical equation
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