

Solution

The experimental findings of Arthur H. Compton reveal that the wavelength of the scattered X-radiation increases than wavelength of incident radiation by an amount $\Delta\lambda$, called the wavelength shift while according to classical physics, the incident and scattered radiation should have the same wavelength. This wavelength shift is in accordance with conservation of energy & linear momentum applied to the photon-electron collisions.

This wavelength shift is given as:

$$\begin{aligned}\Delta\lambda &= \lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta) \\ &= 2\lambda_c \sin^2\left(\frac{\theta}{2}\right) \dots (1)\end{aligned}$$

where λ' is the wavelength of scattered radiation, λ is the wavelength of incident radiation, m_e is the electron rest mass, h is the Planck's constant, c is the speed of light in vacuum, θ is the scattering angle and $\lambda_c = \frac{h}{m_e c} = 2.426 \times 10^{-12} \text{ m}$ is called the Compton wavelength of the electron.

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