

Chapter 4 Atomic Structure

Section 4.2 The Structure of an Atom**(pages 108–112)**

This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.

Reading Strategy (page 108)

Monitoring Your Understanding Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Know About Atoms	What I Would Like to Learn	What I Have Learned
Most students will know that atoms are the “building blocks” of matter, and some may know that atoms contain subatomic particles.	Based on the title of the section, students may say that they want to learn more about the structure of atoms.	

Properties of Subatomic Particles (pages 108–109)

- What are three subatomic particles?
a. Protons b. Electrons c. Neutrons
- Circle the letter that identifies a subatomic particle with a positive charge.
a. nucleus **(b)** proton
c. neutron d. electron
- Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.

Comparing Subatomic Particles (pages 109–110)

- Circle the letters of properties that vary among subatomic particles.
a. color **(b)** mass
(c) charge **(d)** location in the atom
- Circle the letter of the expression that accurately compares the masses of neutrons and protons.
(a) mass of 1 neutron = mass of 1 proton
b. mass of 2000 neutrons = mass of 1 proton
c. mass of 1 electron = mass of 1 proton
d. mass of 1 neutron = mass of 1 electron

Mcdougal Chapter 20 Section 4 Worksheet Answers

Michal Rosen-Zvi



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Whispering the Secrets of Language: An Psychological Journey through **Mcdougal Chapter 20 Section 4 Worksheet Answers**

In a digitally-driven world wherever screens reign supreme and quick conversation drowns out the subtleties of language, the profound secrets and psychological nuances hidden within words often move unheard. However, nestled within the pages of **Mcdougal Chapter 20 Section 4 Worksheet Answers** a charming fictional value blinking with organic feelings, lies an extraordinary quest waiting to be undertaken. Composed by a skilled wordsmith, that charming opus encourages visitors on an introspective trip, lightly unraveling the veiled truths and profound influence resonating within the fabric of every word. Within the emotional depths of this emotional review, we shall embark upon a honest exploration of the book is primary styles, dissect their charming publishing model, and yield to the strong resonance it evokes heavy within the recesses of readers hearts.

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