## Matter—Properties and Changes

#### Lesson 2.1 Properties of Matter

In your textbook, read about physical properties and chemical properties of matter.

Use each of the terms below just once to complete the passage.

| chemical ma         |                           |   | physical                      |                      |  |
|---------------------|---------------------------|---|-------------------------------|----------------------|--|
| density             | prope                     | rties   | substance                     |                      |  |
| Matter is anythin   | ng with (1)               | MASS  | and volume.                   | A                    |  |
| (2) _SUBSTANCE      | I                         | is a form of n  | natter with a uniform and une | changing             |  |
| composition. Substi | ances have speci          | ific, unchanging (3)                                  | PROPERTIES                    | that can be          |  |
| observed. Substance | es have both phy          | vsical and chemical;                                  | properties. (4)               |                      |  |
| PHYSICAL            |                           | propertie   | s can be observed without ch  | anging a substance's |  |
| chemical compositi  | on. Color, hardr          | ness, and   |                               |                      |  |
| (5)DENSITY_         |                           | are example   | s. Other properties cannot be | observed without     |  |
| changing the compo  | sition of a subs          | tance. These are call                                 | ed (6)CHEMICAL                |                      |  |
| properties. An exam | iple is the tender        | ncy of iron to form r                                 | ust when exposed to air.      |                      |  |
| Label each proper   | ty as either <i>pl</i> (y | sical or chemical.                                    |                               |                      |  |
| CHEM                | 7. Cb                     | 7. Chemical formula H <sub>2</sub> O                  |                               |                      |  |
| CHEM                | 8. Fo                     | 8. Forms green carbonate when exposed to moist air    |                               |                      |  |
| CHEM                | 9. Re                     | 9. Remains unchanged when in the presence of nitrogen |                               |                      |  |
| PHYS                | 10.                       | Colorless   |                               |                      |  |
| PHYS                | 11.                       | Solid at norma  | I temperatures and pressures  |                      |  |
| CHEM                | 12.                       | Ability to combine with another substance             |                               |                      |  |
| PHYS                | 13.                       | Melting point   |                               |                      |  |
| PHYS                | 14.                       | Liquid at norm  | al temperatures and pressure  | s                    |  |
| PHYS                | 15.                       | Boiling point it                                      | s 100°C                       |                      |  |

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# **Matter Properties And Changes Study Guide**

**RD Boyd** 

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical Resources for Teaching Middle School Science Smithsonian aspect of the material discussed in the chapter Institution, National Academy of Engineering, National Science Resources Center of the National Academy of Sciences, Institute of Medicine, 1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and

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