

Questionnaire-Chapter 2

Read & Repeat 17. Identify a correct statement by inserting the answer to the question below in the highlighted space. **Repeat** 18. Write a correct statement by inserting the correct description of the reaction in the highlighted space. **Repeat** 19.

17. (2 pts) Complete the reaction above by inserting the following equations:



- Calcium carbonate is the reactant? $\text{CaO}_{(s)} + \text{CO}_{2(g)}$
- Calcium carbonate is the product? $\text{CaCO}_{3(s)}$
- The correct products are CaO and CO_2
- The correct products are CO_2 and CaO

18. (2 pts) Complete the reaction above by inserting the following equations:



- Inserting all coefficients is correct in the reactant?
Insert the answer: ☐ Yes ☒ No
- Inserting all coefficients is correct in the product?
Insert the answer: ☐ Yes ☒ No

19. (2 pts) One mole of CO_2 is the product of the reaction. How many moles of CaO are needed to react with it? What is the total mass of the reactants?

One mole

100.09 g

20. (2 pts) What mass of CaO is needed to react with 1 mole of laughing gas? Is the reaction exothermic? How many moles of gas? How many moles of laughing gas are in the tank?

One mole

Learning Guide To Molar Relationship Answers

**Jo Ann Robinson, Troy Polite, Nancy
Musick**



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