

Aromatic Substitution Reactions

1. The correct systematic name for the aromatic compound shown is



(b) 4-ethyl-1-methylbenzene

(c) 1-ethyl-4-methylbenzene (this may be considered to give the lowest

- (d) lowest number for the lowest-order substituent giving priority to the ethyl group. It is often observed that both (c) and (d) are given with the benzene ring as 1, is used with methyl groups.
(e) there is not a correct systematic (IUPAC) name for this compound.



(b) 4-ethyl-1-methylbenzene is correct (a phenyl group)

2. Name the most common electrophilic aromatic substitution reaction characterizing the aromatic species benzene. Explain its kinetics (what rate law does it obey) and its mechanism (what is the rate-determining step)?

- (a) Electrophilic aromatic substitution
(b) Electrophilic aromatic substitution
(c) Electrophilic aromatic substitution

3. What is the correct systematic name for the aromatic compound shown?



(b) 1,3-dibromobenzene

4. What feature of benzene explains the relatively low reactivity of benzene towards electrophilic aromatic substitution? What is the kinetic effect of this low reactivity?



- (b) The substituent group stabilizes the pi bond system.
(c) Substituent stabilization of pi bonds.

(d) The substituent group stabilizes the pi bond system.

5. Electrophilic aromatic substitution reactions of benzene are characterized by which of the following?

- (a) The most substituted carbon must become the type of compound.
(b) Substitution occurs for the most stable pi bond system.
(c) Substitution occurs for the most stable pi bond system.

6. What is the correct systematic name for the aromatic compound shown?

Learnsmart Chemistry Answers

A Gutmann



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