# Passage 1 (Questions 1-5)

To lead how amono acids first appeared on this planet, an experiment was set up to simulate the conditions. thought to have existed on early Earth. H y O. Off y , NH y , and H y were all introduced into a startle. arrangement of sealed flashs and tubes. Expoil water evaporated in one compartment as "lightning" was induced between a jour of electrodes. The water vigor than occled and condensed in a connected section. which fact back to the first compartment, until a disrains assultation even reached.

After two weeks of continuous operation, roughly 2% of the carbon in the system was in the form of among acids, with glucke being the most abundant. The following steps are believed to account for the formation of the key intermediates, formalitelyde and flydrogen syands:

#### Methadolog 1

Via the Stracker Synthesis, glycine can then form as follows:

#### Mechanism 2

Several other amino acids were also observed to form. Of the chiral amino acids, both erunitoriesy were created in a racernic miniture. Trisugh in most biological systems Liumino acids are used exclusively for proteir synfless.

Simple sugars also formed in the same environment, Once formaldehyde is present to react with H y O. b. does so according to the Bullerov reaction:

Figure 1. Formation of curbohydrates from aqueous formaldwhyde

# Proper

#### Question 4

Which of the following will increase the yould of carbodylic acid formed from a nitria group containing compound. using a mechanism similar to the Strecker synthesis?

- C. A. Use a stronger nucleophile, like a Grignant Reagent, instead of H y C.
- B. Attach an electron withdrawing group to the carton of the nitrile group.
- C. Use a stronger electrophile. The a Gropourd Reaguest, instead of H y O
- O. D. Attach an electron donating group to the carbon of the ritirle group.

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# **Kaplan Mcat Practice Test Answers**

David D. Reid

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