Experiment 8 Report Sheet Limiting Reactant

Date	Lab Sec	Name		Desk No.
A. Precipit	nation of CaC ₃ O ₂ H	O from the Salt Mixture		
Unknown matther			Trial I	Trial 2
L. Mass of Seaker (g)			73.059.0	7043019
2. Mass of beater and salt mixture (g)		75.13.73	71.43240	
3. Mass of salt mexture (g)		1. 5765 9	1.00283	
4. Many of	Mass of filter paper (g)		9.39683	0.39629
5. Mass of filter paper and product after air-dried or swem-dried (c)			9.7609	0.705899
6. Mass of	Mass of dried product (g)		0.36417 9	0.304691
7. Fermula	Formula of dried product		Cacyon	Co C , O ,
B. Determin	nation of Limiting	Reactant	70 17	
1. Limiting	reaction in salt me	sture (write complete formu	(a) K2C2	Q4.40
2. Excess a	eactant in salt most	ure (write complete formula	Coci	12-2H20
Data Analys	ás .			
1. Moles of	Moles of CaC ₂ O ₄ ·H ₂ O (or CaC ₂ O ₄) precipitated (w		2. 496 ×10 3	r 2.120 110 -01
	Moles of limiting reactant in salt mixture (wof) Formula of limiting hydrate Fy Cy Qu		1.980 NO-3	1-681 x100
3. Mass of	class of limiting reactant in salt mixture (g)		0-36479	0.309679
	Mass of excess reactant in salt mixture (g) formula of excess hydrate Co. C.I. O.y.		1.21187	0.69319
5. Percent l	Percent limiting reactant in salt mixture (%)		23.13 %	30. 88 2.
6. Percent	Percent excess reactant in salt mixture (%)		76.87%	69.127.
7. Mass of	Mass of excess reactant that reacted (g)		0.28939	0.2456 9
B. Mass of	Mass of excess reactant, unreacted (g)		0.922 56	0.44759

^{*}Show calculations for Trial 1 on next page,

Laboratory Manual Limiting Reactant

Shasha Hu

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