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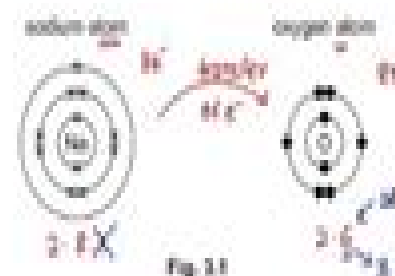


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3 This question is about ionic and covalent compounds. [4 marks: Q₃ → 1M]

- (a) (i) Sodium reacts with oxygen to form the ionic compound sodium oxide. [1M]
The electronic configurations of an atom of sodium and an atom of oxygen are shown in Fig. 3.1.



Ions are formed by the transfer of electrons from sodium atoms to oxygen atoms.

Complete the electron shell diagrams in Fig. 3.2 to show the electronic configuration of one sodium ion and one oxide ion. Show the charges on the ions.



- (ii) Write the formula of sodium oxide.

Na₂O



[2]

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