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WRITING NEUTRAL FORMULAS FOR IONIC COMPOUNDS

Directions: Use a periodic table. Remember that A groups have the following ionic charges: Group 1 (+1), Group 2 (+2), Group 3 (+3), Group 5 (-3), Group 6 (-2), and Group 7 (-1). Transition metals can have variable positive charge, which is given to you in the compound name (ex. fron (IIII) has a +3 charge whereas Copper (II) has a +2 charge).

#	Compound Name	Cation	Anion	Neutral Ionic Formula
1	Sodium chloride	Na⁺	CI-	NaCl
2	Calcium fluoride	Ca ²⁺	F-	CaF ₂
3	Aluminum chloride			
4	Aluminum oxide			
5	Beryllium fluoride			
6	Sodium oxide			
7	Copper (II) sulfide			
8	Aluminum fluoride			
9	Manganese (III) oxide			
10	Magnesium nitride			
11	Sodium sulfide		12	
12	Iron (III) oxide			

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Basil Mugaga Naah

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In some sort of inundated with displays and the cacophony of instantaneous connection, the profound power and mental resonance of verbal art usually disappear into obscurity, eclipsed by the constant onslaught of noise and distractions. However, nestled within the lyrical pages of **Naming And Writing Formulas For Ionic Compound Chapter Worksheet Answers**, a charming work of literary beauty that impulses with raw feelings, lies an remarkable journey waiting to be embarked upon. Composed with a virtuoso wordsmith, that enchanting opus manuals viewers on an emotional odyssey, lightly exposing the latent potential and profound influence embedded within the delicate web of language. Within the heartwrenching expanse of this evocative analysis, we shall embark upon an introspective exploration of the book is central themes, dissect their interesting writing model, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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