

## Polyatomic Ions Worksheet

### Polyatomic Practice

1. Name or write the formula for the following polyatomic ions

|              |       |                     |       |
|--------------|-------|---------------------|-------|
| sulfate      | _____ | $\text{CO}_3^{2-}$  | _____ |
| nitrite      | _____ | $\text{MnO}_4^{1-}$ | _____ |
| perphosphate | _____ | $\text{SO}_3^{2-}$  | _____ |
| hypiodite    | _____ | $\text{BrO}_2^{1-}$ | _____ |
| chlorite     | _____ | $\text{CO}_3^{2-}$  | _____ |
| phosphite    | _____ | $\text{PO}_3^{2-}$  | _____ |
| percarbonate | _____ | $\text{ClO}^{1-}$   | _____ |
| bromate      | _____ | $\text{IO}_3^{1-}$  | _____ |
| hyposulfite  | _____ | $\text{PO}_4^{2-}$  | _____ |
| permanganate | _____ | $\text{NO}_3^{1-}$  | _____ |
| carbonate    | _____ | $\text{SO}_4^{2-}$  | _____ |

2. Name or write the formula for the following Type I polyatomic ionic compounds

|                         |       |   |       |
|-------------------------|-------|---|-------|
| beryllium hydroxide     | _____ | $\text{Ba}(\text{IO}_3)_2$                    | _____ |
| sodium nitrite          | _____ | $\text{Ga}(\text{CNO})_3$                     | _____ |
| ammonium chloride       | _____ | $\text{Ag}_2\text{SO}_4$                      | _____ |
| calcium bisulfate       | _____ | $\text{MgCO}$                                 | _____ |
| rubidium perchlorate    | _____ | $\text{NH}_2\text{NO}_2$                      | _____ |
| strontium sulfite       | _____ | $\text{Al}(\text{C}_2\text{H}_3\text{O}_5)_3$ | _____ |
| aluminum acetate        | _____ | $\text{SrSO}_4$                               | _____ |
| ammonium nitrate        | _____ | $\text{RbClO}_2$                              | _____ |
| magnesium hypocarbonate | _____ | $\text{Ca}(\text{HSO}_4)_2$                   | _____ |
| silver hyposulfite      | _____ | $\text{NH}_4\text{Cl}$                        | _____ |
| gallium cyanate         | _____ | $\text{NaNO}$                                 | _____ |
| barium iodate           | _____ | $\text{Be}(\text{OH})_2$                      | _____ |

# Naming Formulas Binary Polyatomic Worksheet

**S Marginson**



### **Naming Formulas Binary Polyatomic Worksheet:**

**Experimental Chemistry** Robert J. Artz, 1982      Physical Science Carson-Dellosa Publishing, 2015-03-09 Physical Science for grades 5 to 12 is designed to aid in the review and practice of physical science topics Physical Science covers topics such as scientific measurement force and energy matter atoms and elements magnetism and electricity The book includes realistic diagrams and engaging activities to support practice in all areas of physical science The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series is aligned to current science standards      **Chemistry** John S. Phillips, Cheryl Wistrom, 2000      **Addison-Wesley Chemistry**, 1990      Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998

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