

# Naming Polyatomic Ionic Compounds Worksheet

Name the following ions.

- 1)  $(\text{NH}_4)_2\text{CrO}_4$  \_\_\_\_\_
- 2)  $\text{AgNO}_3$  \_\_\_\_\_
- 3)  $\text{Al}(\text{ClO}_3)_3$  \_\_\_\_\_
- 4)  $\text{Ba}(\text{ClO}_4)_2$  \_\_\_\_\_
- 5)  $\text{CaSO}_4$  \_\_\_\_\_
- 6)  $\text{Cd}(\text{OH})_2$  \_\_\_\_\_
- 7)  $\text{Cu}(\text{ClO}_4)_2$  \_\_\_\_\_
- 8)  $\text{Fe}(\text{HSO}_3)_2$  \_\_\_\_\_
- 9)  $\text{Fe}(\text{NO}_2)_2$  \_\_\_\_\_
- 10)  $\text{Hg}_2(\text{NO}_3)_2$  \_\_\_\_\_
- 11)  $\text{HgCrO}_4$  \_\_\_\_\_
- 12)  $\text{K}_2\text{Cr}_2\text{O}_7$  \_\_\_\_\_
- 13)  $\text{Mg}_3(\text{PO}_4)_2$  \_\_\_\_\_
- 14)  $\text{Pb}(\text{CO}_3)_2$  \_\_\_\_\_
- 15)  $\text{Pb}(\text{SO}_4)_2$  \_\_\_\_\_
- 16)  $\text{Pb}_3(\text{PO}_4)_2$  \_\_\_\_\_
- 17)  $\text{Sn}(\text{CN})_2$  \_\_\_\_\_
- 18)  $\text{Sn}(\text{CO}_3)_2$  \_\_\_\_\_
- 19)  $\text{Sr}(\text{NO}_3)_2$  \_\_\_\_\_
- 20)  $\text{Zn}(\text{HCO}_3)_2$  \_\_\_\_\_

# Naming Ionic Compounds With Polyatomic Ions Answers

**Susan McMurry**



## **Naming Ionic Compounds With Polyatomic Ions Answers:**

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**eBook: General, Organic and Biological Chemistry 2e** SMITH, 2012-02-16  
**eBook General Organic and Biological Chemistry 2e** **Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

**General, Organic, and Biological Chemistry Study Guide and Selected Solutions** Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **Study Guide and Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry, Third Edition** John McMurry, Susan McMurry, 1999 Provides worked out solutions to text problems along with chapter by chapter outlines and a variety of self

tests at the end of each chapter      **Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry** Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana Sue Baker Publishing News      **Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today** Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application      Basic Chemistry Steven S. Zumdahl, 1999 For a full description see catalog entry for Zumdahl Introductory Chemistry A Foundation 4 e      Chemistry Karen C. Timberlake, 1992      **Introductory Chemistry** Steven S. Zumdahl, 1996      *Chemistry Demystified* Linda Williams, 2003-06-26 Say goodbye to dry presentations grueling formulas and abstract theory that would put Einstein to sleep now there s an easier way to master chemistry biology trigonometry and geometry McGraw Hill s Demystified Series teaches complex subjects in a unique easy to absorb manner and is designed for users without formal training unlimited time or genius IQs Organized like self teaching guides they come complete with key points background information questions at the end of each chapter and final exams There s no better way to gain instant expertise ABOUT CHEMISTRY DEMYSTIFIED Current real world examples illustrate the essential nature of the basic elements as they form various states of gases liquids and solids Covers essentials such as understanding matter chemical building blocks elements electrons and the periodic table properties and reactions Includes unique Chembits tips scientific news and cutting edge industry applications not usually found in textbooks      *Partial Solutions Manual* Darrell D. Ebbing, David Bookin, 1998      Physical Science - Chemistry Split with Online Learning Center Password Card (Chapters 1 And 8 - 13) Bill W. Tillery, Tillery Bill, 2004-01-09      *Prentice Hall Chemistry*, 2000      **Students Study Guide** Fredeen, Pearson, 2003-07      Introductory Chemistry Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking      *Chemistry* Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives      **Chemistry with Inorganic Qualitative Analysis** Therald Moeller, 1984      **Introduction to Chemistry, Study Guide** T. R. Dickson, 1999-07-07 Teaches chemistry by

offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition has been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images     Integrated Science Bill W. Tillery,Eldon D. Enger,Frederick C. Ross,2001 Integrated Science is an easy to read but substantial introduction to the fundamental behavior of matter and energy in living and nonliving systems It is intended to serve the needs of non science majors who are required to complete one or more science courses as part of a general or basic studies requirement It introduces basic concepts and key ideas while providing opportunities for students to learn reasoning skills and a new way of thinking about their environment No prior work in science is assumed The language as well as the mathematics is as simple as can be practical for a college level science course

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