

# Naming Acids Worksheet Packet

Marrie: Aynsuler Key

1. What is the definition of an acid?

An acid is a substance that when in water gives off hydrogen tons

2. What are the rules for naming acids?

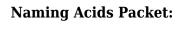
If no oxygen in anion HYDRO + root of anion + IC acid.
If axygen add - IC to root of anion or element if -are ending on union or add -OUS to root of anion or element if -its ending on anion.

3. Write the names and the unions for the following acids.

Acid H <sub>5</sub> SO <sub>4</sub>	Anion	Name sulfuric acid
H <sub>2</sub> PO <sub>1</sub>	phosphate	phosphoric acid
$HC_2H_4O_2$	acetate:	acetic acid
H <sub>2</sub> SO <sub>5</sub>	sulfilico	sulfurous acid
HNO <sub>2</sub>	miterite	nitrous acid
HCIO <sub>4</sub>	perchlorate	perchloric acid
HCIO <sub>3</sub>	chilorate	ebforie acid
HCIO <sub>2</sub>	chlorite	chlorous acid
HCIO	hypochlorite	hypochlorous acid

# **Naming Acids Packet**

**David Baud** 



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