



# Nomenclature for polyatomic ions Worksheet-Answer Key

**Exercise 1.** Complete the table of neutral ionic compounds with the formulas and names for each cation-anion pair, similar to Exercise 2.

	$\text{NH}_4^+$	$\text{Na}^+$	$\text{PO}_4^{3-}$	$\text{CO}_3^{2-}$	$\text{ClO}_2^-$	$\text{OH}^-$
$\text{Na}^+$	$\text{Na}_2\text{SO}_4$ sodium sulfate	$\text{NaNO}_3$ sodium nitrate	$\text{Na}_3\text{PO}_4$ sodium phosphate	$\text{Na}_2\text{CO}_3$ sodium carbonate	$\text{NaClO}_2$ sodium chlorite	$\text{NaOH}$ sodium hydroxide
$\text{Al}^{3+}$	$\text{Al}_2(\text{SO}_4)_3$ aluminum sulfate	$\text{AlNO}_3$ aluminum nitrate	$\text{AlPO}_4$ aluminum phosphate	$\text{Al}_2(\text{CO}_3)_3$ aluminum carbonate	$\text{AlClO}_2$ aluminum chlorite	$\text{AlOH}$ aluminum hydroxide
$\text{Ba}^{2+}$	$\text{BaSO}_4$ barium sulfate	$\text{Ba}(\text{NO}_3)_2$ barium nitrate	$\text{Ba}_3(\text{PO}_4)_2$ barium phosphate	$\text{BaCO}_3$ barium carbonate	$\text{BaClO}_2$ barium chlorite	$\text{BaOH}$ barium hydroxide
$\text{NH}_4^+$	$(\text{NH}_4)_2\text{SO}_4$ ammonium sulfate	$\text{NH}_4\text{NO}_3$ ammonium nitrate	$(\text{NH}_4)_3\text{PO}_4$ ammonium phosphate	$(\text{NH}_4)_2\text{CO}_3$ ammonium carbonate	$\text{NH}_4\text{ClO}_2$ ammonium chlorite	$\text{NH}_4\text{OH}$ ammonium hydroxide
$\text{Ca}^{2+}$	$\text{Ca}_2\text{SO}_4$ calcium sulfate	$\text{CaNO}_3$ calcium nitrate	$\text{Ca}_3\text{PO}_4$ calcium phosphate	$\text{Ca}_2\text{CO}_3$ calcium carbonate	$\text{CaClO}_2$ calcium chlorite	$\text{CaOH}$ calcium hydroxide

# Naming Polyatomic Ions Packet Answer Key

**Paul B. Kelter, Steven S. Zumdahl**



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