

## The Hydrogen Atom – Student Guide

### Background Material

Carefully read the background pages entitled Energy Levels, Light, and Transitions and answer the following questions to check your understanding.

Question 1: Complete the following table which compares how the Bohr Model and the Quantum model represent the Hydrogen atom. In some cases they both describe things in the same way and in some cases they do not.

Concept	Bohr Model	Quantum Model
Nucleus	positively charged proton at the center	positively charged proton at the center
Electrons		negatively charged cloud
Orbital		
Location of electron in the first orbital	always located at the Bohr radius	
Energy of the Energy Levels		$E = -E_0/n^2$

Question 2: Complete the following table which relates the parameters of two different photons by circling the appropriate relationship. The first row is completed for you: "A red photon has a larger wavelength, smaller frequency, smaller energy, and the same velocity through space as a blue photon".

# Naap Hydrogen Atom Lab Answer Key

**Karin Nielsen-Saines**



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## Whispering the Strategies of Language: An Psychological Journey through **Naap Hydrogen Atom Lab Answer Key**

In a digitally-driven world where displays reign great and quick transmission drowns out the subtleties of language, the profound strategies and emotional nuances hidden within phrases frequently move unheard. Yet, situated within the pages of **Naap Hydrogen Atom Lab Answer Key** a interesting literary treasure pulsing with organic thoughts, lies a fantastic quest waiting to be undertaken. Composed by an experienced wordsmith, that marvelous opus attracts viewers on an introspective trip, softly unraveling the veiled truths and profound impact resonating within the very material of each word. Within the psychological depths with this touching evaluation, we can embark upon a heartfelt exploration of the book is core styles, dissect their captivating publishing model, and succumb to the effective resonance it evokes serious within the recesses of readers hearts.

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