NAMING IONIC AND COMMON COVALENT COMPOUNDS

1)	Write the name of the following ionic compounds. For the transition element			
	compounds, you must include the appropriate valency. These are marked with an *			
	For example, MnO ₂ is named manganese (IV) oxide.			

a)	NaOH	73	
ь)	Ca ₃ (PO ₄) ₂		
c)	MgO	50 <u></u>	
d)	Al ₂ S ₃	54	
e)	K ₂ CO ₃		
	Li ₂ SO ₃		
	BaCl ₂	5.	
h)	NH ₄ CI	F	

2) Write the names of the following covalent compounds and/or common acids.

a)	SO ₃			
b)	H ₂ SO ₄	1		
C)	NH ₃	# }		
	co			
	CCl4			
f)	СН₃СООН			
g)	H ₂ O	-		
h)	CH ₄	g <u></u>		
i)	H₃PO₄	30 [
D	NO ₂	-0.		

D *CuSO4

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S Ben Porath

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