



CHEMISTRY

CHEAT SHEET



NAMING SIMPLE COMPOUNDS

NAMING SIMPLE COMPOUNDS CHEAT SHEET

Naming Compounds with Their Formulas
 (Show the ions involved in the compound)



Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.
5. If the compound is a molecular compound, use prefixes to indicate the number of atoms of each element.
6. If the compound is an acid, use the prefix H- followed by the name of the anion.

Binary Compounds

First Element	Anion
H	hydrogen
C	carbon
N	nitrogen
O	oxygen
F	fluorine
Cl	chlorine
S	sulfur
Br	bromine
I	iodine
Al	aluminum
Fe	iron
Cu	copper
Zn	zinc
Ag	silver
Au	gold

Common Ions and Their Names

Formula	Name
Na^+	Sodium
K^+	Potassium
Ca^{2+}	Calcium
Mg^{2+}	Magnesium
Al^{3+}	Aluminum
Fe^{2+}	Iron (II)
Fe^{3+}	Iron (III)
Cu^+	Copper (I)
Cu^{2+}	Copper (II)
Zn^{2+}	Zinc
Ag^+	Silver
Au^+	Gold (I)
Au^{3+}	Gold (III)

Naming Compounds with a Central and Peripheral



Sodium Chloride

Directions for naming:

1. Write the cation (metal) name first.
2. Write the anion (non-metal) name second, changing the ending to -ide.
3. If the cation is a transition metal, include the oxidation state in Roman numerals in parentheses.
4. If the anion is a polyatomic ion, use its name.

Naming Acids

Formula	Central Element	Formula	Central Element
HCl	Hydrogen	HNO_3	Nitrogen
HBr	Bromine	H_2SO_4	Sulfur
HI	Iodine	H_3PO_4	Phosphorus
H_2S	Sulfur	H_2CO_3	Carbon
H_2SO_4	Sulfur	H_2SiO_3	Silicon
HNO_3	Nitrogen	H_2SiO_4	Silicon

Common Names for Specific Compounds

Formula	Common Name	Formula	Common Name
NaCl	Table salt	H_2O	Water
HCl	Hydrochloric acid	CO_2	Carbon dioxide
H_2SO_4	Sulfuric acid	CH_4	Methane
HNO_3	Nitric acid	C_2H_6	Ethane

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Reviewing **Naming Compounds Cheat Sheet**: Unlocking the Spellbinding Force of Linguistics

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